

# Carbonic acid, but-3-en-1-yl heptadecyl ester

**Inchi:** InChI=1S/C22H42O3/c1-3-5-7-8-9-10-11-12-13-14-15-16-17-18-19-21-25-22(23)24-20-6  
**InchiKey:** IZKMGIHXJFVCKF-UHFFFAOYSA-N  
**Formula:** C22H42O3  
**SMILES:** C=CCCOC(=O)OCCCCCCCCCCCCCCCCCC  
**Mol. weight [g/mol]:** 354.57

## Physical Properties

Property code	Value	Unit	Source
gf	-116.72	kJ/mol	Joback Method
hf	-749.00	kJ/mol	Joback Method
hfus	55.43	kJ/mol	Joback Method
hvap	75.46	kJ/mol	Joback Method
log10ws	-7.81		Crippen Method
logp	7.587		Crippen Method
mvol	329.850	ml/mol	McGowan Method
pc	941.52	kPa	Joback Method
rinpol	2412.00		NIST Webbook
rinpol	2412.00		NIST Webbook
tb	798.15	K	Joback Method
tc	978.42	K	Joback Method
tf	430.33	K	Joback Method
vc	1.290	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1037.68	J/molxK	798.15	Joback Method
cpg	1057.58	J/molxK	828.19	Joback Method
cpg	1076.42	J/molxK	858.24	Joback Method
cpg	1094.23	J/molxK	888.28	Joback Method
cpg	1111.04	J/molxK	918.33	Joback Method
cpg	1126.87	J/molxK	948.37	Joback Method
cpg	1141.73	J/molxK	978.42	Joback Method
dvisc	0.0009012	Paxs	430.33	Joback Method

dvisc	0.0003995	Paxs	491.63	Joback Method
dvisc	0.0002121	Paxs	552.94	Joback Method
dvisc	0.0001278	Paxs	614.24	Joback Method
dvisc	0.0000844	Paxs	675.54	Joback Method
dvisc	0.0000597	Paxs	736.85	Joback Method
dvisc	0.0000446	Paxs	798.15	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U383238&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U383238&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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