

# Fumaric acid, 2-ethylbutyl dodec-2-en-1-yl ester

Inchi:	InChI=1S/C22H38O4/c1-4-7-8-9-10-11-12-13-14-15-18-25-21(23)16-17-22(24)26-19-20(
InchiKey:	DITXYFPIAZACAO-JLXBFWJWSA-N
Formula:	C22H38O4
SMILES:	CCCCCCCCC=CCOC(=O)C=CC(=O)OCC(CC)CC
Mol. weight [g/mol]:	366.53

## Physical Properties

Property code	Value	Unit	Source
gf	-175.48	kJ/mol	Joback Method
hf	-757.85	kJ/mol	Joback Method
hfus	55.19	kJ/mol	Joback Method
hvap	82.41	kJ/mol	Joback Method
log10ws	-6.22		Crippen Method
logp	5.762		Crippen Method
mcvol	327.120	ml/mol	McGowan Method
pc	1023.34	kPa	Joback Method
rinpol	2544.00		NIST Webbook
tb	863.22	K	Joback Method
tc	1058.48	K	Joback Method
tf	456.86	K	Joback Method
vc	1.270	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1039.88	J/molxK	863.22	Joback Method
cpg	1057.91	J/molxK	895.76	Joback Method
cpg	1074.88	J/molxK	928.31	Joback Method
cpg	1090.83	J/molxK	960.85	Joback Method
cpg	1105.82	J/molxK	993.40	Joback Method
cpg	1119.87	J/molxK	1025.94	Joback Method
cpg	1133.05	J/molxK	1058.48	Joback Method
dvisc	0.0006816	Paxs	456.86	Joback Method
dvisc	0.0002831	Paxs	524.59	Joback Method

dvisc	0.0001438	Paxs	592.31	Joback Method
dvisc	0.0000839	Paxs	660.04	Joback Method
dvisc	0.0000541	Paxs	727.77	Joback Method
dvisc	0.0000376	Paxs	795.49	Joback Method
dvisc	0.0000277	Paxs	863.22	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405641&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405641&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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