

# 1,2-Cyclohexanedicarboxylic acid, 3,5-dimethylcyclohexyl hexyl ester

Inchi:	InChI=1S/C22H38O4/c1-4-5-6-9-12-25-21(23)19-10-7-8-11-20(19)22(24)26-18-14-16(2)
InchiKey:	WEYNQVXJYNZXOX-UHFFFAOYSA-N
Formula:	C22H38O4
SMILES:	CCCCCOC(=O)C1CCCCC1C(=O)OC1CC(C)CC(C)C1
Mol. weight [g/mol]:	366.53

## Physical Properties

Property code	Value	Unit	Source
gf	-307.71	kJ/mol	Joback Method
hf	-939.39	kJ/mol	Joback Method
hfus	45.19	kJ/mol	Joback Method
hvap	82.81	kJ/mol	Joback Method
log10ws	-5.69		Crippen Method
logp	5.284		Crippen Method
mvol	314.000	ml/mol	McGowan Method
pc	1167.22	kPa	Joback Method
rinpol	2467.00		NIST Webbook
rinpol	2467.00		NIST Webbook
tb	880.43	K	Joback Method
tc	1092.67	K	Joback Method
tf	484.06	K	Joback Method
vc	1.179	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1097.46	J/molxK	880.43	Joback Method
cpg	1118.04	J/molxK	915.80	Joback Method
cpg	1136.74	J/molxK	951.18	Joback Method
cpg	1153.58	J/molxK	986.55	Joback Method
cpg	1168.57	J/molxK	1021.92	Joback Method
cpg	1181.72	J/molxK	1057.30	Joback Method
cpg	1193.05	J/molxK	1092.67	Joback Method
dvisc	0.0010786	Paxs	484.06	Joback Method

dvisc	0.0005625	Paxs	550.12	Joback Method
dvisc	0.0003373	Paxs	616.18	Joback Method
dvisc	0.0002233	Paxs	682.25	Joback Method
dvisc	0.0001590	Paxs	748.31	Joback Method
dvisc	0.0001197	Paxs	814.37	Joback Method
dvisc	0.0000940	Paxs	880.43	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339850&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339850&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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