

# Diglycolic acid, di(2-isopropylphenyl) ester

<b>Inchi:</b>	InChI=1S/C22H26O5/c1-15(2)17-9-5-7-11-19(17)26-21(23)13-25-14-22(24)27-20-12-8-6
<b>InchiKey:</b>	UZAPBJLVRRZRNM-UHFFFAOYSA-N
<b>Formula:</b>	C22H26O5
<b>SMILES:</b>	CC(C)c1ccccc1OC(=O)COCC(=O)Oc1ccccc1C(C)C
<b>Mol. weight [g/mol]:</b>	370.44

## Physical Properties

Property code	Value	Unit	Source
gf	-237.80	kJ/mol	Joback Method
hf	-679.67	kJ/mol	Joback Method
hfus	39.76	kJ/mol	Joback Method
hvap	90.39	kJ/mol	Joback Method
log10ws	-5.21		Crippen Method
logp	4.461		Crippen Method
mcvol	294.070	ml/mol	McGowan Method
pc	1471.36	kPa	Joback Method
rinpol	3157.00		NIST Webbook
rinpol	3157.00		NIST Webbook
tb	940.20	K	Joback Method
tc	1167.22	K	Joback Method
tf	552.13	K	Joback Method
vc	1.105	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	925.14	J/molxK	940.20	Joback Method
cpg	938.67	J/molxK	978.04	Joback Method
cpg	950.66	J/molxK	1015.87	Joback Method
cpg	961.13	J/molxK	1053.71	Joback Method
cpg	970.11	J/molxK	1091.54	Joback Method
cpg	977.62	J/molxK	1129.38	Joback Method
cpg	983.68	J/molxK	1167.22	Joback Method
dvisc	0.0003123	Paxs	552.13	Joback Method

dvisc	0.0001675	Paxs	616.81	Joback Method
dvisc	0.0001011	Paxs	681.49	Joback Method
dvisc	0.0000666	Paxs	746.16	Joback Method
dvisc	0.0000469	Paxs	810.84	Joback Method
dvisc	0.0000348	Paxs	875.52	Joback Method
dvisc	0.0000269	Paxs	940.20	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382294&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382294&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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