

# Glutaric acid, 2-methylpent-3-yl (2-naphthyl)methyl ester

**Inchi:** InChI=1S/C22H28O4/c1-4-20(16(2)3)26-22(24)11-7-10-21(23)25-15-17-12-13-18-8-5-6-9  
**InchiKey:** LPRZGQMFKFWSBQ-UHFFFAOYSA-N  
**Formula:** C22H28O4  
**SMILES:** CCC(OC(=O)CCCC(=O)OCc1ccc2ccccc2c1)C(C)C  
**Mol. weight [g/mol]:** 356.46

## Physical Properties

Property code	Value	Unit	Source
gf	-128.93	kJ/mol	Joback Method
hf	-581.44	kJ/mol	Joback Method
hfus	41.93	kJ/mol	Joback Method
hvap	86.68	kJ/mol	Joback Method
log10ws	-6.36		Crippen Method
logp	5.031		Crippen Method
mcvol	292.500	ml/mol	McGowan Method
pc	1412.25	kPa	Joback Method
rinpol	2754.00		NIST Webbook
rinpol	2754.00		NIST Webbook
tb	905.10	K	Joback Method
tc	1122.68	K	Joback Method
tf	523.66	K	Joback Method
vc	1.117	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	920.99	J/molxK	905.10	Joback Method
cpg	936.11	J/molxK	941.36	Joback Method
cpg	950.05	J/molxK	977.63	Joback Method
cpg	962.88	J/molxK	1013.89	Joback Method
cpg	974.66	J/molxK	1050.15	Joback Method
cpg	985.43	J/molxK	1086.42	Joback Method
cpg	995.27	J/molxK	1122.68	Joback Method
dvisc	0.0007057	Paxs	523.66	Joback Method

dvisc	0.0003892	Paxs	587.23	Joback Method
dvisc	0.0002411	Paxs	650.81	Joback Method
dvisc	0.0001626	Paxs	714.38	Joback Method
dvisc	0.0001170	Paxs	777.95	Joback Method
dvisc	0.0000885	Paxs	841.53	Joback Method
dvisc	0.0000696	Paxs	905.10	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392197&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392197&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/87-071-7/Glutaric-acid-2-methylpent-3-yl-2-naphthyl-methyl-ester.pdf>

Generated by Cheméo on 2024-05-02 01:37:25.257672275 +0000 UTC m=+16903094.178249591.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.