

# 1,2-Cyclohexanedicarboxylic acid, isohexyl 4-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C21H30O5/c1-15(2)7-6-14-25-20(22)18-8-4-5-9-19(18)21(23)26-17-12-10-16(2)
<b>InchiKey:</b>	XGFOFQIRAQSMNY-UHFFFAOYSA-N
<b>Formula:</b>	C21H30O5
<b>SMILES:</b>	COc1ccc(OC(=O)C2CCCCC2C(=O)OCCCC(C)C)cc1
<b>Mol. weight [g/mol]:</b>	362.46

## Physical Properties

Property code	Value	Unit	Source
gf	-329.82	kJ/mol	Joback Method
hf	-844.83	kJ/mol	Joback Method
hfus	39.94	kJ/mol	Joback Method
hvap	85.73	kJ/mol	Joback Method
log10ws	-4.96		Crippen Method
logp	4.386		Crippen Method
mcvol	292.880	ml/mol	McGowan Method
pc	1410.13	kPa	Joback Method
rinpol	2656.00		NIST Webbook
rinpol	2656.00		NIST Webbook
tb	900.98	K	Joback Method
tc	1121.17	K	Joback Method
tf	520.06	K	Joback Method
vc	1.095	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	966.41	J/molxK	900.98	Joback Method
cpg	982.90	J/molxK	937.68	Joback Method
cpg	997.67	J/molxK	974.38	Joback Method
cpg	1010.75	J/molxK	1011.08	Joback Method
cpg	1022.14	J/molxK	1047.78	Joback Method
cpg	1031.85	J/molxK	1084.47	Joback Method
cpg	1039.90	J/molxK	1121.17	Joback Method
dvisc	0.0005225	Paxs	520.06	Joback Method

dvisc	0.0002765	Paxs	583.55	Joback Method
dvisc	0.0001657	Paxs	647.03	Joback Method
dvisc	0.0001089	Paxs	710.52	Joback Method
dvisc	0.0000766	Paxs	774.01	Joback Method
dvisc	0.0000569	Paxs	837.49	Joback Method
dvisc	0.0000440	Paxs	900.98	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339670&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339670&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/87-156-3/1-2-Cyclohexanedicarboxylic-acid-isohehexyl-4-methoxyphenyl-ester.pdf>

Generated by Cheméo on 2024-04-23 15:56:05.275117667 +0000 UTC m=+16177014.195694978.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.