

# Glutaric acid, 2-ethylhexyl 4-acetylphenyl ester

Inchi:	InChI=1S/C21H30O5/c1-4-6-8-17(5-2)15-25-20(23)9-7-10-21(24)26-19-13-11-18(12-14-
InchiKey:	PHYGVXYMVQJHQW-UHFFFAOYSA-N
Formula:	C21H30O5
SMILES:	CCCCC(CC)COC(=O)CCCC(=O)Oc1ccc(C(C)=O)cc1
Mol. weight [g/mol]:	362.46

## Physical Properties

Property code	Value	Unit	Source
gf	-370.48	kJ/mol	Joback Method
hf	-859.17	kJ/mol	Joback Method
hfus	47.45	kJ/mol	Joback Method
hvap	89.95	kJ/mol	Joback Method
log10ws	-5.67		Crippen Method
logp	4.724		Crippen Method
mcvol	299.440	ml/mol	McGowan Method
pc	1317.52	kPa	Joback Method
rinpola	2732.00		NIST Webbook
rinpola	2732.00		NIST Webbook
tb	917.55	K	Joback Method
tc	1128.93	K	Joback Method
tf	544.62	K	Joback Method
vc	1.151	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	953.73	J/molxK	917.55	Joback Method
cpg	968.28	J/molxK	952.78	Joback Method
cpg	981.54	J/molxK	988.01	Joback Method
cpg	993.55	J/molxK	1023.24	Joback Method
cpg	1004.33	J/molxK	1058.47	Joback Method
cpg	1013.91	J/molxK	1093.70	Joback Method
cpg	1022.32	J/molxK	1128.93	Joback Method
dvisc	0.0004831	Paxs	544.62	Joback Method

dvisc	0.0002602	Paxs	606.77	Joback Method
dvisc	0.0001572	Paxs	668.93	Joback Method
dvisc	0.0001035	Paxs	731.09	Joback Method
dvisc	0.0000727	Paxs	793.24	Joback Method
dvisc	0.0000538	Paxs	855.39	Joback Method
dvisc	0.0000415	Paxs	917.55	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392035&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392035&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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