

Undec-10-ynoic acid, 2,7-dimethyloct-1-en-3-yn-5-yl ester

Inchi: InChI=1S/C21H32O2/c1-6-7-8-9-10-11-12-13-14-21(22)23-20(17-19(4)5)16-15-18(2)3/h1-5,11-14,17-19,21-23H,20H2,22H3
InchiKey: FLSHKOLQAWMBKK-UHFFFAOYSA-N
Formula: C21H32O2
SMILES: C#CCCCCCCCC(=O)OC(C#CC(=C)C)CC(C)C
Mol. weight [g/mol]: 316.48

Physical Properties

Property code	Value	Unit	Source
gf	392.30	kJ/mol	Joback Method
hf	-52.29	kJ/mol	Joback Method
hfus	49.39	kJ/mol	Joback Method
hvap	72.14	kJ/mol	Joback Method
log10ws	-6.79		Crippen Method
logp	5.278		Crippen Method
mcvol	292.690	ml/mol	McGowan Method
pc	1260.16	kPa	Joback Method
rinpol	2090.00		NIST Webbook
rinpol	2090.00		NIST Webbook
tb	750.97	K	Joback Method
tc	945.73	K	Joback Method
tf	505.94	K	Joback Method
vc	1.129	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	844.69	J/mol×K	750.97	Joback Method
cpg	863.31	J/mol×K	783.43	Joback Method
cpg	880.93	J/mol×K	815.89	Joback Method
cpg	897.59	J/mol×K	848.35	Joback Method
cpg	913.31	J/mol×K	880.81	Joback Method
cpg	928.15	J/mol×K	913.27	Joback Method
cpg	942.14	J/mol×K	945.73	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U406967&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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