

# 1-Phenanthrenecarboxylic acid, 7-ethenyl-1,2,3,4,4a,5,6,7,8,9,10,10a-dodecahydro-

Other names: Podocarp-8-en-15-oic acid, 13«alpha»-methyl-13-vinyl-, methyl ester

methyl ester Methyl 8,15-Pimaradien-18-oate

**[1R-(1«alpha»,4a«beta»,7«beta»,10a«alpha»)]-**

InchiKey: GTFNGUBYWRFHMR-UHFFFAOYSA-N

Formula: C<sub>21</sub>H<sub>32</sub>O<sub>2</sub>

SMILES: C=CC1(C)CCC2=C(CCC3C(C)(C(=O)OC)CCCC23C)C1

Mol. weight [g/mol]: 316.48

CAS: 3582-26-1

## Physical Properties

Property code	Value	Unit	Source
gf	88.13	kJ/mol	Joback Method
hf	-348.32	kJ/mol	Joback Method
hfus	18.18	kJ/mol	Joback Method
hvap	69.28	kJ/mol	Joback Method
log10ws	-5.90		Crippen Method
logp	5.439		Crippen Method
mcvol	273.010	ml/mol	McGowan Method
pc	1570.96	kPa	Joback Method
rinpol	2184.00		NIST Webbook
rinpol	2184.00		NIST Webbook
tb	799.59	K	Joback Method
tc	1036.90	K	Joback Method
tf	526.31	K	Joback Method
vc	1.026	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	881.99	J/mol×K	799.59	Joback Method
cpg	908.01	J/mol×K	839.14	Joback Method
cpg	934.29	J/mol×K	878.69	Joback Method
cpg	961.27	J/mol×K	918.24	Joback Method
cpg	989.38	J/mol×K	957.79	Joback Method

cpg	1019.06	J/mol×K	997.34	Joback Method
cpg	1050.73	J/mol×K	1036.90	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3582261&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3582261&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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