

# Isophthalic acid, 1-naphthyl propyl ester

<b>Inchi:</b>	InChI=1S/C21H18O4/c1-2-13-24-20(22)16-9-5-10-17(14-16)21(23)25-19-12-6-8-15-7-3-4
<b>InchiKey:</b>	DAVSXTYHXUCMHT-UHFFFAOYSA-N
<b>Formula:</b>	C21H18O4
<b>SMILES:</b>	CCCOC(=O)c1cccc(C(=O)Oc2cccc3ccccc23)c1
<b>Mol. weight [g/mol]:</b>	334.37

## Physical Properties

Property code	Value	Unit	Source
gf	-29.69	kJ/mol	Joback Method
hf	-325.18	kJ/mol	Joback Method
hfus	40.04	kJ/mol	Joback Method
hvap	88.17	kJ/mol	Joback Method
log10ws	-6.44		Crippen Method
logp	4.626		Crippen Method
mvol	254.650	ml/mol	McGowan Method
pc	1949.23	kPa	Joback Method
rinpol	2969.00		NIST Webbook
rinpol	2969.00		NIST Webbook
tb	914.76	K	Joback Method
tc	1155.83	K	Joback Method
tf	581.33	K	Joback Method
vc	0.966	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	758.45	J/molxK	914.76	Joback Method
cpg	771.10	J/molxK	954.94	Joback Method
cpg	782.55	J/molxK	995.12	Joback Method
cpg	792.88	J/molxK	1035.30	Joback Method
cpg	802.17	J/molxK	1075.47	Joback Method
cpg	810.48	J/molxK	1115.65	Joback Method
cpg	817.90	J/molxK	1155.83	Joback Method
dvisc	0.0005468	Paxs	581.33	Joback Method

dvisc	0.0003651	Paxs	636.90	Joback Method
dvisc	0.0002601	Paxs	692.47	Joback Method
dvisc	0.0001949	Paxs	748.05	Joback Method
dvisc	0.0001519	Paxs	803.62	Joback Method
dvisc	0.0001224	Paxs	859.19	Joback Method
dvisc	0.0001012	Paxs	914.76	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344686&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344686&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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