

# Dimethylmalonic acid, nonyl trans-4-methylcyclohexyl ester

Inchi:	InChI=1S/C21H38O4/c1-5-6-7-8-9-10-11-16-24-19(22)21(3,4)20(23)25-18-14-12-17(2)13
InchiKey:	JIQNHCVCCQGNRL-UHFFFAOYSA-N
Formula:	C21H38O4
SMILES:	CCCCCCCCCOC(=O)C(C)(C)C(=O)OC1CCC(C)CC1
Mol. weight [g/mol]:	354.52

## Physical Properties

Property code	Value	Unit	Source
gf	-322.32	kJ/mol	Joback Method
hf	-941.14	kJ/mol	Joback Method
hfus	41.21	kJ/mol	Joback Method
hvap	79.48	kJ/mol	Joback Method
log10ws	-5.86		Crippen Method
logp	5.428		Crippen Method
mvol	310.770	ml/mol	McGowan Method
pc	1148.32	kPa	Joback Method
rinpol	2304.00		NIST Webbook
rinpol	2304.00		NIST Webbook
tb	844.11	K	Joback Method
tc	1044.78	K	Joback Method
tf	476.31	K	Joback Method
vc	1.181	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1031.55	J/molxK	844.11	Joback Method
cpg	1115.64	J/molxK	1011.33	Joback Method
cpg	1101.47	J/molxK	977.89	Joback Method
cpg	1086.01	J/molxK	944.44	Joback Method
cpg	1069.23	J/molxK	911.00	Joback Method
cpg	1051.09	J/molxK	877.55	Joback Method
cpg	1128.56	J/molxK	1044.78	Joback Method
dvisc	0.0000486	Paxs	844.11	Joback Method

dvisc	0.0000650	Paxs	782.81	Joback Method
dvisc	0.0000914	Paxs	721.51	Joback Method
dvisc	0.0001369	Paxs	660.21	Joback Method
dvisc	0.0002229	Paxs	598.91	Joback Method
dvisc	0.0004053	Paxs	537.61	Joback Method
dvisc	0.0008599	Paxs	476.31	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363901&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363901&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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