

# 2,4-Dichlorobenzyl alcohol, n-butyl ether

<b>Other names:</b>	2,4-dichlorobenzyl butyl ether
<b>Inchi:</b>	InChI=1S/C11H14Cl2O/c1-2-3-6-14-8-9-4-5-10(12)7-11(9)13/h4-5,7H,2-3,6,8H2,1H3
<b>InchiKey:</b>	JSCVZXJIQORJNU-UHFFFAOYSA-N
<b>Formula:</b>	C11H14Cl2O
<b>SMILES:</b>	CCCCOCc1ccc(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	233.13

## Physical Properties

Property code	Value	Unit	Source
gf	6.03	kJ/mol	Joback Method
hf	-220.48	kJ/mol	Joback Method
hfus	27.09	kJ/mol	Joback Method
hvap	54.86	kJ/mol	Joback Method
log10ws	-4.48		Crippen Method
logp	4.310		Crippen Method
mcvol	172.440	ml/mol	McGowan Method
pc	2347.36	kPa	Joback Method
rinpol	1605.00		NIST Webbook
rinpol	1569.00		NIST Webbook
rinpol	1559.00		NIST Webbook
rinpol	1605.00		NIST Webbook
rinpol	1561.00		NIST Webbook
rinpol	1562.00		NIST Webbook
rinpol	1563.00		NIST Webbook
rinpol	1566.00		NIST Webbook
rinpol	1569.00		NIST Webbook
rinpol	1568.00		NIST Webbook
rinpol	1569.00		NIST Webbook
rinpol	1572.00		NIST Webbook
rinpol	1555.00		NIST Webbook
rinpol	1559.00		NIST Webbook
rinpol	1562.00		NIST Webbook
rinpol	1563.00		NIST Webbook
tb	585.00	K	Joback Method
tc	796.84	K	Joback Method
tf	347.26	K	Joback Method
vc	0.659	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.96	J/molxK	585.00	Joback Method
cpg	442.12	J/molxK	761.53	Joback Method
cpg	431.48	J/molxK	726.23	Joback Method
cpg	420.17	J/molxK	690.92	Joback Method
cpg	408.15	J/molxK	655.61	Joback Method
cpg	395.42	J/molxK	620.31	Joback Method
cpg	452.09	J/molxK	796.84	Joback Method
dvisc	0.0001713	Paxs	585.00	Joback Method
dvisc	0.0002127	Paxs	545.38	Joback Method
dvisc	0.0002732	Paxs	505.75	Joback Method
dvisc	0.0003661	Paxs	466.13	Joback Method
dvisc	0.0005181	Paxs	426.51	Joback Method
dvisc	0.0007871	Paxs	386.88	Joback Method
dvisc	0.0013156	Paxs	347.26	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378106&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378106&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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