

# 9-Octadecene, 1,1-dimethoxy-, (Z)-

<b>Other names:</b>	Olealdehyde, dimethyl acetal
<b>Inchi:</b>	InChI=1S/C20H40O2/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20(21-2)22-3/h11-1
<b>InchiKey:</b>	WBNMCUYSZJIEFY-QXMHVHEDSA-N
<b>Formula:</b>	C20H40O2
<b>SMILES:</b>	CCCCCCCCC=CCCCCCCCC(OC)OC
<b>Mol. weight [g/mol]:</b>	312.53
<b>CAS:</b>	15677-71-1

## Physical Properties

Property code	Value	Unit	Source
gf	-14.70	kJ/mol	Joback Method
hf	-608.63	kJ/mol	Joback Method
hfus	46.61	kJ/mol	Joback Method
hvap	64.50	kJ/mol	Joback Method
log10ws	-6.83		Crippen Method
logp	6.643		Crippen Method
mcvol	300.100	ml/mol	McGowan Method
pc	1032.57	kPa	Joback Method
rinpol	2374.00		NIST Webbook
rinpol	2374.00		NIST Webbook
tb	705.56	K	Joback Method
tc	874.47	K	Joback Method
tf	339.54	K	Joback Method
vc	1.165	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	887.29	J/molxK	705.56	Joback Method
cpg	907.59	J/molxK	733.71	Joback Method
cpg	926.98	J/molxK	761.86	Joback Method
cpg	945.50	J/molxK	790.01	Joback Method
cpg	963.16	J/molxK	818.16	Joback Method
cpg	979.99	J/molxK	846.32	Joback Method

cpg	996.00	J/molxK	874.47	Joback Method
dvisc	0.0019341	Paxs	339.54	Joback Method
dvisc	0.0006440	Paxs	400.54	Joback Method
dvisc	0.0002867	Paxs	461.55	Joback Method
dvisc	0.0001542	Paxs	522.55	Joback Method
dvisc	0.0000944	Paxs	583.55	Joback Method
dvisc	0.0000634	Paxs	644.56	Joback Method
dvisc	0.0000457	Paxs	705.56	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15677711&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15677711&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/87-306-6/9-Octadecene-1-1-dimethoxy-Z.pdf>

Generated by Cheméo on 2024-04-26 09:31:25.770010092 +0000 UTC m=+16413134.690587408.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.