

# Pentadecanoic acid, 2,6,10,14-tetramethyl-, methyl ester

**Other names:**

Pentadecanoic acid, 2,6,10,14-tetramethyl-, methyl ester, (2R,6R,10R)-  
2,6,10,14-Tetramethylpentadecanoic acid, methyl ester  
Methyl nonadecanoate (isoprenoid)  
Pristanic acid, methyl ester

**Inchi:**

InChI=1S/C20H40O2/c1-16(2)10-7-11-17(3)12-8-13-18(4)14-9-15-19(5)20(21)22-6/h16-1

**InchiKey:**

LCWYDFQCGCPILG-UHFFFAOYSA-N

**Formula:**

C20H40O2

**SMILES:**

COC(=O)C(C)CCCC(C)CCCC(C)CCCC(C)C

**Mol. weight [g/mol]:**

312.53

**CAS:**

1001-80-5

## Physical Properties

Property code	Value	Unit	Source
gf	-126.16	kJ/mol	Joback Method
hf	-722.05	kJ/mol	Joback Method
hfus	36.25	kJ/mol	Joback Method
hvap	67.72	kJ/mol	Joback Method
log10ws	-6.09		Crippen Method
logp	6.235		Crippen Method
mcvol	300.100	ml/mol	McGowan Method
pc	1067.97	kPa	Joback Method
tb	731.53	K	Joback Method
tc	908.43	K	Joback Method
tf	327.32	K	Joback Method
vc	1.155	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	910.24	J/mol×K	731.53	Joback Method
cpg	1002.33	J/mol×K	878.94	Joback Method
cpg	985.83	J/mol×K	849.46	Joback Method
cpg	968.39	J/mol×K	819.98	Joback Method
cpg	950.00	J/mol×K	790.50	Joback Method

cpg	930.62	J/mol×K	761.01	Joback Method
cpg	1017.92	J/mol×K	908.43	Joback Method
dvisc	0.0000532	Paxs	731.53	Joback Method
dvisc	0.0000775	Paxs	664.16	Joback Method
dvisc	0.0001229	Paxs	596.79	Joback Method
dvisc	0.0002192	Paxs	529.42	Joback Method
dvisc	0.0004628	Paxs	462.06	Joback Method
dvisc	0.0012608	Paxs	394.69	Joback Method
dvisc	0.0051894	Paxs	327.32	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1001805&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1001805&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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