

# 4,4'-((2R,3S)-2,3-Dimethylbutane-1,4-diyl)bis(2-me

<b>Inchi:</b>	InChI=1S/C20H26O4/c1-13(9-15-5-7-17(21)19(11-15)23-3)14(2)10-16-6-8-18(22)20(12-
<b>InchiKey:</b>	ADFOLUXMYYCTRR-UHFFFAOYSA-N
<b>Formula:</b>	C20H26O4
<b>SMILES:</b>	COc1cc(CC(C)C(C)Cc2ccc(O)c(OC)c2)ccc1O
<b>Mol. weight [g/mol]:</b>	330.42
<b>CAS:</b>	66322-34-7

## Physical Properties

Property code	Value	Unit	Source
gf	-201.04	kJ/mol	Joback Method
hf	-635.63	kJ/mol	Joback Method
hfus	41.76	kJ/mol	Joback Method
hvap	96.06	kJ/mol	Joback Method
log10ws	-4.42		Crippen Method
logp	4.172		Crippen Method
mvol	268.620	ml/mol	McGowan Method
pc	2018.13	kPa	Joback Method
rinpol	2714.60		NIST Webbook
rinpol	2714.60		NIST Webbook
tb	925.52	K	Joback Method
tc	1160.51	K	Joback Method
tf	630.94	K	Joback Method
vc	0.895	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	860.72	J/molxK	925.52	Joback Method
cpg	937.29	J/molxK	1121.34	Joback Method
cpg	922.52	J/molxK	1082.18	Joback Method
cpg	907.62	J/molxK	1043.01	Joback Method
cpg	892.44	J/molxK	1003.85	Joback Method
cpg	876.86	J/molxK	964.68	Joback Method
cpg	952.05	J/molxK	1160.51	Joback Method

dvisc	8.0544022e-08	Paxs	925.52	Joback Method
dvisc	0.0000001	Paxs	876.42	Joback Method
dvisc	0.0000002	Paxs	827.33	Joback Method
dvisc	0.0000004	Paxs	778.23	Joback Method
dvisc	0.0000007	Paxs	729.13	Joback Method
dvisc	0.0000015	Paxs	680.04	Joback Method
dvisc	0.0000035	Paxs	630.94	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C66322347&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C66322347&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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