

# Fumaric acid, 4-phenoxybenzyl propyl ester

<b>Inchi:</b>	InChI=1S/C20H20O5/c1-2-14-23-19(21)12-13-20(22)24-15-16-8-10-18(11-9-16)25-17-6-
<b>InchiKey:</b>	IECBZSDCEITCPF-OUKQBFOZSA-N
<b>Formula:</b>	C20H20O5
<b>SMILES:</b>	CCCOC(=O)C=CC(=O)OCc1ccc(Oc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	340.37

## Physical Properties

Property code	Value	Unit	Source
gf	-159.91	kJ/mol	Joback Method
hf	-499.14	kJ/mol	Joback Method
hfus	42.21	kJ/mol	Joback Method
hvap	86.01	kJ/mol	Joback Method
log10ws	-4.49		Crippen Method
logp	4.031		Crippen Method
mvol	261.590	ml/mol	McGowan Method
pc	1789.39	kPa	Joback Method
rinpol	2606.00		NIST Webbook
tb	894.50	K	Joback Method
tc	1123.16	K	Joback Method
tf	541.99	K	Joback Method
vc	0.986	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	780.88	J/molxK	894.50	Joback Method
cpg	833.88	J/molxK	1085.05	Joback Method
cpg	825.70	J/molxK	1046.94	Joback Method
cpg	816.35	J/molxK	1008.83	Joback Method
cpg	805.79	J/molxK	970.72	Joback Method
cpg	793.98	J/molxK	932.61	Joback Method
cpg	840.93	J/molxK	1123.16	Joback Method
dvisc	0.0000368	Paxs	894.50	Joback Method
dvisc	0.0000468	Paxs	835.75	Joback Method

dvisc	0.0000616	Paxs	777.00	Joback Method
dvisc	0.0000849	Paxs	718.25	Joback Method
dvisc	0.0001239	Paxs	659.49	Joback Method
dvisc	0.0001946	Paxs	600.74	Joback Method
dvisc	0.0003370	Paxs	541.99	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348110&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348110&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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