

# Fumaric acid, di(3-ethylphenyl) ester

**Inchi:** InChI=1S/C20H20O4/c1-3-15-7-5-9-17(13-15)23-19(21)11-12-20(22)24-18-10-6-8-16(4-2)  
**InchiKey:** SGPKQMANPYGMQT-VAWYXSNFSA-N  
**Formula:** C20H20O4  
**SMILES:** CCc1cccc(OC(=O)C=CC(=O)Oc2cccc(CC)c2)c1  
**Mol. weight [g/mol]:** 324.37

## Physical Properties

Property code	Value	Unit	Source
gf	-64.54	kJ/mol	Joback Method
hf	-378.39	kJ/mol	Joback Method
hfus	40.64	kJ/mol	Joback Method
hvap	84.26	kJ/mol	Joback Method
log10ws	-5.21		Crippen Method
logp	3.879		Crippen Method
mvol	255.720	ml/mol	McGowan Method
pc	1792.42	kPa	Joback Method
rinpol	2634.00		NIST Webbook
rinpol	2634.00		NIST Webbook
tb	877.06	K	Joback Method
tc	1107.75	K	Joback Method
tf	532.28	K	Joback Method
vc	0.968	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	752.87	J/molxK	877.06	Joback Method
cpg	766.57	J/molxK	915.51	Joback Method
cpg	779.05	J/molxK	953.96	Joback Method
cpg	790.37	J/molxK	992.41	Joback Method
cpg	800.57	J/molxK	1030.85	Joback Method
cpg	809.71	J/molxK	1069.30	Joback Method
cpg	817.84	J/molxK	1107.75	Joback Method
dvisc	0.0004152	Paxs	532.28	Joback Method

dvisc	0.0002455	Paxs	589.74	Joback Method
dvisc	0.0001593	Paxs	647.21	Joback Method
dvisc	0.0001110	Paxs	704.67	Joback Method
dvisc	0.0000816	Paxs	762.13	Joback Method
dvisc	0.0000627	Paxs	819.60	Joback Method
dvisc	0.0000498	Paxs	877.06	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348161&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348161&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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