

# Fumaric acid, butyl 4-phenylphenyl ester

<b>Inchi:</b>	InChI=1S/C20H20O4/c1-2-3-15-23-19(21)13-14-20(22)24-18-11-9-17(10-12-18)16-7-5-4
<b>InchiKey:</b>	KYBLIMZAEQYWIQ-BUHFOSPRSA-N
<b>Formula:</b>	C20H20O4
<b>SMILES:</b>	CCCCOC(=O)C=CC(=O)Oc1ccc(-c2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	324.37

## Physical Properties

Property code	Value	Unit	Source
gf	-54.91	kJ/mol	Joback Method
hf	-366.92	kJ/mol	Joback Method
hfus	41.02	kJ/mol	Joback Method
hvap	83.60	kJ/mol	Joback Method
log10ws	-5.62		Crippen Method
logp	4.158		Crippen Method
mcvol	255.720	ml/mol	McGowan Method
pc	1815.41	kPa	Joback Method
rinpol	2726.00		NIST Webbook
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tb	872.08	K	Joback Method
tc	1102.04	K	Joback Method
tf	519.76	K	Joback Method
vc	0.968	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	754.10	J/molxK	872.08	Joback Method
cpg	811.53	J/molxK	1063.71	Joback Method
cpg	802.24	J/molxK	1025.39	Joback Method
cpg	791.90	J/molxK	987.06	Joback Method
cpg	780.48	J/molxK	948.73	Joback Method
cpg	767.90	J/molxK	910.41	Joback Method
cpg	819.85	J/molxK	1102.04	Joback Method
dvisc	0.0000490	Paxs	872.08	Joback Method

dvisc	0.0000624	Paxs	813.36	Joback Method
dvisc	0.0000826	Paxs	754.64	Joback Method
dvisc	0.0001145	Paxs	695.92	Joback Method
dvisc	0.0001685	Paxs	637.20	Joback Method
dvisc	0.0002684	Paxs	578.48	Joback Method
dvisc	0.0004750	Paxs	519.76	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348209&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348209&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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