

# Benzoic acid, 4-[(butoxycarbonyl)oxy]-, 4-ethoxyphenyl ester

<b>Other names:</b>	Butyl p-(p-ethoxyphenoxycarbonyl)-phenyl carbonate Butyl p-(ethoxyphenoxycarbonyl)phenyl carbonate Butyl p-(p-ethoxyphenyloxycarbonyl)phenyl carbonate Butyl para-(para-ethoxyphenoxycarbonyl)phenylcarbonate BEPC Carbonic acid, butyl ester, ester with p-ethoxyphenyl p-hydroxybenzoate 4-ethoxyphenyl 4-[(butoxycarbonyl)oxy]benzoate
<b>Inchi:</b>	InChI=1S/C20H22O6/c1-3-5-14-24-20(22)26-18-8-6-15(7-9-18)19(21)25-17-12-10-16(11)
<b>InchiKey:</b>	GIJIKFKGLZGENN-UHFFFAOYSA-N
<b>Formula:</b>	C20H22O6
<b>SMILES:</b>	CCCCOC(=O)Oc1ccc(C(=O)Oc2ccc(OCC)cc2)cc1
<b>Mol. weight [g/mol]:</b>	358.39
<b>CAS:</b>	16494-24-9

## Physical Properties

Property code	Value	Unit	Source
gf	-354.76	kJ/mol	Joback Method
hf	-760.05	kJ/mol	Joback Method
hfus	42.81	kJ/mol	Joback Method
hvap	89.12	kJ/mol	Joback Method
log10ws	-5.73		Crippen Method
logp	4.620		Crippen Method
mcvol	271.760	ml/mol	McGowan Method
pc	1668.70	kPa	Joback Method
tb	917.74	K	Joback Method
tc	1141.32	K	Joback Method
tf	581.82	K	Joback Method
vc	1.024	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	834.35	J/mol×K	917.74	Joback Method
cpg	882.11	J/mol×K	1104.06	Joback Method

cpg	875.56	J/molxK	1066.79	Joback Method
cpg	867.50	J/molxK	1029.53	Joback Method
cpg	857.95	J/molxK	992.27	Joback Method
cpg	846.90	J/molxK	955.00	Joback Method
cpg	887.16	J/molxK	1141.32	Joback Method
dvisc	0.0000321	Paxs	917.74	Joback Method
dvisc	0.0000401	Paxs	861.75	Joback Method
dvisc	0.0000516	Paxs	805.77	Joback Method
dvisc	0.0000691	Paxs	749.78	Joback Method
dvisc	0.0000968	Paxs	693.79	Joback Method
dvisc	0.0001441	Paxs	637.81	Joback Method
dvisc	0.0002313	Paxs	581.82	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16494249&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16494249&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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