

# 9H-Fluorene-9-carboxylic acid, hexyl ester

**Inchi:** InChI=1S/C20H22O2/c1-2-3-4-9-14-22-20(21)19-17-12-7-5-10-15(17)16-11-6-8-13-18(16)  
**InchiKey:** HOXNWWCBJKARBH-UHFFFAOYSA-N  
**Formula:** C20H22O2  
**SMILES:** CCCCCCOC(=O)C1c2ccccc2-c2ccccc21  
**Mol. weight [g/mol]:** 294.39

## Physical Properties

Property code	Value	Unit	Source
gf	174.11	kJ/mol	Joback Method
hf	-165.69	kJ/mol	Joback Method
hfus	39.98	kJ/mol	Joback Method
hvap	74.72	kJ/mol	Joback Method
log10ws	-6.18		Crippen Method
logp	4.922		Crippen Method
mvol	241.720	ml/mol	McGowan Method
pc	1765.41	kPa	Joback Method
rinpol	3216.00		NIST Webbook
rinpol	3216.00		NIST Webbook
tb	794.81	K	Joback Method
tc	1015.20	K	Joback Method
tf	490.18	K	Joback Method
vc	0.936	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	713.75	J/molxK	794.81	Joback Method
cpg	784.64	J/molxK	978.46	Joback Method
cpg	772.07	J/molxK	941.73	Joback Method
cpg	758.79	J/molxK	905.00	Joback Method
cpg	744.71	J/molxK	868.27	Joback Method
cpg	729.73	J/molxK	831.54	Joback Method
cpg	796.59	J/molxK	1015.20	Joback Method
dvisc	0.0004567	Paxs	794.81	Joback Method

dvisc	0.0005232	Paxs	744.04	Joback Method
dvisc	0.0006114	Paxs	693.27	Joback Method
dvisc	0.0007324	Paxs	642.50	Joback Method
dvisc	0.0009049	Paxs	591.72	Joback Method
dvisc	0.0011633	Paxs	540.95	Joback Method
dvisc	0.0015753	Paxs	490.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415129&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415129&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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