

# Diglycolic acid, di(phenethyl) ester

<b>Inchi:</b>	InChI=1S/C20H22O5/c21-19(24-13-11-17-7-3-1-4-8-17)15-23-16-20(22)25-14-12-18-9-5
<b>InchiKey:</b>	FWZJIESNNCLMAL-UHFFFAOYSA-N
<b>Formula:</b>	C20H22O5
<b>SMILES:</b>	O=C(COCC(=O)OCCc1ccccc1)OCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	342.39

## Physical Properties

Property code	Value	Unit	Source
gf	-230.50	kJ/mol	Joback Method
hf	-604.89	kJ/mol	Joback Method
hfus	42.40	kJ/mol	Joback Method
hvap	85.39	kJ/mol	Joback Method
log10ws	-3.21		Crippen Method
logp	2.575		Crippen Method
mvol	265.890	ml/mol	McGowan Method
pc	1734.67	kPa	Joback Method
rinpol	3444.00		NIST Webbook
rinpol	3444.00		NIST Webbook
tb	885.36	K	Joback Method
tc	1107.59	K	Joback Method
tf	534.55	K	Joback Method
vc	1.006	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	809.91	J/molxK	885.36	Joback Method
cpg	823.50	J/molxK	922.40	Joback Method
cpg	835.74	J/molxK	959.44	Joback Method
cpg	846.66	J/molxK	996.48	Joback Method
cpg	856.29	J/molxK	1033.51	Joback Method
cpg	864.66	J/molxK	1070.55	Joback Method
cpg	871.79	J/molxK	1107.59	Joback Method
dvisc	0.0004184	Paxs	534.55	Joback Method

dvisc	0.0002354	Paxs	593.02	Joback Method
dvisc	0.0001468	Paxs	651.49	Joback Method
dvisc	0.0000990	Paxs	709.96	Joback Method
dvisc	0.0000709	Paxs	768.42	Joback Method
dvisc	0.0000532	Paxs	826.89	Joback Method
dvisc	0.0000414	Paxs	885.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382168&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382168&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/87-402-9/Diglycolic-acid-di-phenethyl-ester.pdf>

Generated by Cheméo on 2024-04-20 13:54:12.053621837 +0000 UTC m=+15910500.974199147.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.