

# 5,5'-((2R,3S)-2,3-Dimethylbutane-1,4-diyl)bis(benz

<b>Inchi:</b>	InChI=1S/C20H22O4/c1-13(7-15-3-5-17-19(9-15)23-11-21-17)14(2)8-16-4-6-18-20(10-16)
<b>InchiKey:</b>	QEFJURUMSHPMTC-UHFFFAOYSA-N
<b>Formula:</b>	C20H22O4
<b>SMILES:</b>	CC(Cc1ccc2c(c1)OCO2)C(C)Cc1ccc2c(c1)OCO2
<b>Mol. weight [g/mol]:</b>	326.39
<b>CAS:</b>	110269-50-6

## Physical Properties

Property code	Value	Unit	Source
gf	91.38	kJ/mol	Joback Method
hf	-381.23	kJ/mol	Joback Method
hfus	53.08	kJ/mol	Joback Method
hvap	85.02	kJ/mol	Joback Method
log10ws	-5.50		Crippen Method
logp	4.201		Crippen Method
mcvol	246.900	ml/mol	McGowan Method
pc	2010.90	kPa	Joback Method
rinpol	2598.30		NIST Webbook
rinpol	2598.30		NIST Webbook
tb	860.02	K	Joback Method
tc	1101.32	K	Joback Method
tf	538.72	K	Joback Method
vc	0.927	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	776.12	J/molxK	860.02	Joback Method
cpg	847.12	J/molxK	1061.11	Joback Method
cpg	834.16	J/molxK	1020.89	Joback Method
cpg	820.73	J/molxK	980.67	Joback Method
cpg	806.69	J/molxK	940.45	Joback Method
cpg	791.87	J/molxK	900.24	Joback Method
cpg	859.78	J/molxK	1101.32	Joback Method

dvisc	0.0002992	Paxs	860.02	Joback Method
dvisc	0.0003630	Paxs	806.47	Joback Method
dvisc	0.0004526	Paxs	752.92	Joback Method
dvisc	0.0005838	Paxs	699.37	Joback Method
dvisc	0.0007854	Paxs	645.82	Joback Method
dvisc	0.0011148	Paxs	592.27	Joback Method
dvisc	0.0016967	Paxs	538.72	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C110269506&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C110269506&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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