

# Fumaric acid, 2-phenethyl 2-ethylhexyl ester

<b>Inchi:</b>	InChI=1S/C20H28O4/c1-3-5-9-17(4-2)16-24-20(22)13-12-19(21)23-15-14-18-10-7-6-8-1
<b>InchiKey:</b>	DDSRMVJVDCAOEG-OUKQBFOZSA-N
<b>Formula:</b>	C20H28O4
<b>SMILES:</b>	CCCCC(CC)COC(=O)C=CC(=O)OCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	332.43

## Physical Properties

Property code	Value	Unit	Source
gf	-160.13	kJ/mol	Joback Method
hf	-597.26	kJ/mol	Joback Method
hfus	43.85	kJ/mol	Joback Method
hvap	80.27	kJ/mol	Joback Method
log10ws	-4.63		Crippen Method
logp	4.088		Crippen Method
mvol	279.480	ml/mol	McGowan Method
pc	1418.64	kPa	Joback Method
rinpol	2401.00		NIST Webbook
rinpol	2401.00		NIST Webbook
tb	839.98	K	Joback Method
tc	1045.27	K	Joback Method
tf	465.82	K	Joback Method
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	852.53	J/molxK	839.98	Joback Method
cpg	868.41	J/molxK	874.19	Joback Method
cpg	883.19	J/molxK	908.41	Joback Method
cpg	896.90	J/molxK	942.62	Joback Method
cpg	909.59	J/molxK	976.84	Joback Method
cpg	921.29	J/molxK	1011.05	Joback Method
cpg	932.05	J/molxK	1045.27	Joback Method
dvisc	0.0007457	Paxs	465.82	Joback Method

dvisc	0.0003506	Paxs	528.18	Joback Method
dvisc	0.0001933	Paxs	590.54	Joback Method
dvisc	0.0001194	Paxs	652.90	Joback Method
dvisc	0.0000802	Paxs	715.26	Joback Method
dvisc	0.0000575	Paxs	777.62	Joback Method
dvisc	0.0000432	Paxs	839.98	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405681&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405681&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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