

# Succinic acid, cyclohexylmethyl 2-isopropoxyphenyl ester

**Inchi:** InChI=1S/C20H28O5/c1-15(2)24-17-10-6-7-11-18(17)25-20(22)13-12-19(21)23-14-16-8-  
**InchiKey:** VVCBVOFOZWGGQY-UHFFFAOYSA-N  
**Formula:** C20H28O5  
**SMILES:** CC(C)Oc1ccccc1OC(=O)CCC(=O)OCC1CCCCC1  
**Mol. weight [g/mol]:** 348.43

## Physical Properties

Property code	Value	Unit	Source
gf	-330.53	kJ/mol	Joback Method
hf	-803.85	kJ/mol	Joback Method
hfus	36.28	kJ/mol	Joback Method
hvap	83.81	kJ/mol	Joback Method
log10ws	-5.14		Crippen Method
logp	4.283		Crippen Method
mvol	278.790	ml/mol	McGowan Method
pc	1557.35	kPa	Joback Method
rinpol	2553.00		NIST Webbook
rinpol	2553.00		NIST Webbook
tb	882.77	K	Joback Method
tc	1104.02	K	Joback Method
tf	513.03	K	Joback Method
vc	1.040	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	902.51	J/molxK	882.77	Joback Method
cpg	968.00	J/molxK	1067.15	Joback Method
cpg	958.06	J/molxK	1030.27	Joback Method
cpg	946.56	J/molxK	993.40	Joback Method
cpg	933.48	J/molxK	956.52	Joback Method
cpg	918.80	J/molxK	919.65	Joback Method
cpg	976.38	J/molxK	1104.02	Joback Method
dvisc	0.0000388	Paxs	882.77	Joback Method

dvisc	0.0000509	Paxs	821.15	Joback Method
dvisc	0.0000699	Paxs	759.52	Joback Method
dvisc	0.0001013	Paxs	697.90	Joback Method
dvisc	0.0001579	Paxs	636.28	Joback Method
dvisc	0.0002706	Paxs	574.65	Joback Method
dvisc	0.0005278	Paxs	513.03	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389792&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389792&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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