

# Icosane-5,7-dione

<b>Inchi:</b>	InChI=1S/C20H38O2/c1-3-5-7-8-9-10-11-12-13-14-15-17-20(22)18-19(21)16-6-4-2/h3-18
<b>InchiKey:</b>	CJHBQGYRPMEB-UHFFFAOYSA-N
<b>Formula:</b>	C20H38O2
<b>SMILES:</b>	CCCCCCCCCCCCC(=O)CC(=O)CCCC
<b>Mol. weight [g/mol]:</b>	310.51

## Physical Properties

Property code	Value	Unit	Source
gf	-140.32	kJ/mol	Joback Method
hf	-681.29	kJ/mol	Joback Method
hfus	50.75	kJ/mol	Joback Method
hvap	73.61	kJ/mol	Joback Method
log10ws	-6.75		Crippen Method
logp	6.406		Crippen Method
mcvol	295.800	ml/mol	McGowan Method
pc	1102.28	kPa	Joback Method
rinpol	2277.60		NIST Webbook
rinpol	2277.60		NIST Webbook
tb	764.74	K	Joback Method
tc	942.91	K	Joback Method
tf	415.02	K	Joback Method
vc	1.167	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	905.39	J/molxK	764.74	Joback Method
cpg	989.55	J/molxK	913.21	Joback Method
cpg	974.45	J/molxK	883.52	Joback Method
cpg	958.52	J/molxK	853.82	Joback Method
cpg	941.72	J/molxK	824.13	Joback Method
cpg	924.02	J/molxK	794.43	Joback Method
cpg	1003.84	J/molxK	942.91	Joback Method
dvisc	0.0000939	Paxs	764.74	Joback Method

dvisc	0.0001254	Paxs	706.45	Joback Method
dvisc	0.0001766	Paxs	648.17	Joback Method
dvisc	0.0002659	Paxs	589.88	Joback Method
dvisc	0.0004381	Paxs	531.59	Joback Method
dvisc	0.0008163	Paxs	473.31	Joback Method
dvisc	0.0018113	Paxs	415.02	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U413683&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U413683&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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