

# Decanoic anhydride

<b>Other names:</b>	n-Capric anhydride Capric anhydride Decanoic acid, anhydride
<b>Inchi:</b>	InChI=1S/C20H38O3/c1-3-5-7-9-11-13-15-17-19(21)23-20(22)18-16-14-12-10-8-6-4-2/h3
<b>InchiKey:</b>	HTWWKYKIBSHDPC-UHFFFAOYSA-N
<b>Formula:</b>	C20H38O3
<b>SMILES:</b>	CCCCCCCCC(=O)OC(=O)CCCCCCCCC
<b>Mol. weight [g/mol]:</b>	326.51
<b>CAS:</b>	2082-76-0

## Physical Properties

Property code	Value	Unit	Source
gf	-245.32	kJ/mol	Joback Method
hf	-813.51	kJ/mol	Joback Method
hfus	51.94	kJ/mol	Joback Method
hvap	76.02	kJ/mol	Joback Method
log10ws	-6.83		Crippen Method
logp	6.338		Crippen Method
mcvol	301.670	ml/mol	McGowan Method
pc	1089.94	kPa	Joback Method
rinpol	2278.00		NIST Webbook
rinpol	2278.00		NIST Webbook
tb	787.16	K	Joback Method
tc	967.84	K	Joback Method
tf	437.25	K	Joback Method
vc	1.185	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	937.35	J/molxK	787.16	Joback Method
cpg	955.79	J/molxK	817.27	Joback Method
cpg	973.26	J/molxK	847.39	Joback Method
cpg	989.78	J/molxK	877.50	Joback Method

cpg	1005.37	J/mol×K	907.62	Joback Method
cpg	1020.06	J/mol×K	937.73	Joback Method
cpg	1033.87	J/mol×K	967.84	Joback Method
dvisc	0.0012318	Paxs	437.25	Joback Method
dvisc	0.0005785	Paxs	495.57	Joback Method
dvisc	0.0003186	Paxs	553.89	Joback Method
dvisc	0.0001965	Paxs	612.20	Joback Method
dvisc	0.0001319	Paxs	670.52	Joback Method
dvisc	0.0000943	Paxs	728.84	Joback Method
dvisc	0.0000709	Paxs	787.16	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2082760&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2082760&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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