

# Pimelic acid, 5-methoxy-3-methylpent-2-yl 4-methyl-2-pentyl ester

Inchi:	InChI=1S/C20H38O5/c1-15(2)14-17(4)24-19(21)10-8-7-9-11-20(22)25-18(5)16(3)12-13-2
InchiKey:	HMGRUGQLTQFHOZ-UHFFFAOYSA-N
Formula:	C20H38O5
SMILES:	COCCC(C)C(C)OC(=O)CCCCC(=O)OC(C)CC(C)C
Mol. weight [g/mol]:	358.51

## Physical Properties

Property code	Value	Unit	Source
gf	-465.08	kJ/mol	Joback Method
hf	-1099.07	kJ/mol	Joback Method
hfus	40.23	kJ/mol	Joback Method
hvap	79.28	kJ/mol	Joback Method
log10ws	-4.75		Crippen Method
logp	4.519		Crippen Method
mvol	313.410	ml/mol	McGowan Method
pc	1088.50	kPa	Joback Method
rinpol	2221.00		NIST Webbook
tb	830.24	K	Joback Method
tc	1020.18	K	Joback Method
tf	421.71	K	Joback Method
vc	1.198	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1002.76	J/molxK	830.24	Joback Method
cpg	1021.03	J/molxK	861.90	Joback Method
cpg	1038.10	J/molxK	893.55	Joback Method
cpg	1053.99	J/molxK	925.21	Joback Method
cpg	1068.69	J/molxK	956.87	Joback Method
cpg	1082.22	J/molxK	988.53	Joback Method
cpg	1094.59	J/molxK	1020.18	Joback Method
dvisc	0.0011834	Paxs	421.71	Joback Method
dvisc	0.0004116	Paxs	489.80	Joback Method

dvisc	0.0001853	Paxs	557.89	Joback Method
dvisc	0.0000992	Paxs	625.98	Joback Method
dvisc	0.0000600	Paxs	694.06	Joback Method
dvisc	0.0000398	Paxs	762.15	Joback Method
dvisc	0.0000282	Paxs	830.24	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406718&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406718&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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