

# 4-Oxo-4-phenylbutyric acid, decyl ester

<b>Inchi:</b>	InChI=1S/C20H30O3/c1-2-3-4-5-6-7-8-12-17-23-20(22)16-15-19(21)18-13-10-9-11-14-18
<b>InchiKey:</b>	CTBKEQYDHQVHKU-UHFFFAOYSA-N
<b>Formula:</b>	C20H30O3
<b>SMILES:</b>	CCCCCCCCCOC(=O)CCC(=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	318.45

## Physical Properties

Property code	Value	Unit	Source
gf	-132.91	kJ/mol	Joback Method
hf	-576.98	kJ/mol	Joback Method
hfus	45.98	kJ/mol	Joback Method
hvap	78.29	kJ/mol	Joback Method
log10ws	-6.02		Crippen Method
logp	5.333		Crippen Method
mvol	277.910	ml/mol	McGowan Method
pc	1373.78	kPa	Joback Method
rinpol	2510.00		NIST Webbook
rinpol	2510.00		NIST Webbook
tb	813.84	K	Joback Method
tc	1012.02	K	Joback Method
tf	463.67	K	Joback Method
vc	1.077	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	848.69	J/molxK	813.84	Joback Method
cpg	865.34	J/molxK	846.87	Joback Method
cpg	880.92	J/molxK	879.90	Joback Method
cpg	895.47	J/molxK	912.93	Joback Method
cpg	909.03	J/molxK	945.96	Joback Method
cpg	921.63	J/molxK	978.99	Joback Method
cpg	933.31	J/molxK	1012.02	Joback Method
dvisc	0.0010056	Paxs	463.67	Joback Method

dvisc	0.0005068	Paxs	522.03	Joback Method
dvisc	0.0002932	Paxs	580.39	Joback Method
dvisc	0.0001875	Paxs	638.75	Joback Method
dvisc	0.0001292	Paxs	697.12	Joback Method
dvisc	0.0000943	Paxs	755.48	Joback Method
dvisc	0.0000720	Paxs	813.84	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405981&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405981&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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