

# 1,2-Cyclohexanedicarboxylic acid, cyclohexylmethyl pentyl ester

**Inchi:** InChI=1S/C20H34O4/c1-2-3-9-14-23-19(21)17-12-7-8-13-18(17)20(22)24-15-16-10-5-4-6  
**InchiKey:** WDTGTPBLLDQGHP-UHFFFAOYSA-N  
**Formula:** C20H34O4  
**SMILES:** CCCCCOC(=O)C1CCCCC1C(=O)OCC1CCCCC1  
**Mol. weight [g/mol]:** 338.48

## Physical Properties

Property code	Value	Unit	Source
gf	-309.13	kJ/mol	Joback Method
hf	-857.43	kJ/mol	Joback Method
hfus	37.87	kJ/mol	Joback Method
hvap	78.97	kJ/mol	Joback Method
log10ws	-4.98		Crippen Method
logp	4.650		Crippen Method
mvol	285.820	ml/mol	McGowan Method
pc	1403.79	kPa	Joback Method
rinpol	2399.00		NIST Webbook
rinpol	2399.00		NIST Webbook
tb	844.01	K	Joback Method
tc	1057.86	K	Joback Method
tf	470.00	K	Joback Method
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	966.74	J/molxK	844.01	Joback Method
cpg	987.02	J/molxK	879.65	Joback Method
cpg	1005.59	J/molxK	915.29	Joback Method
cpg	1022.47	J/molxK	950.94	Joback Method
cpg	1037.69	J/molxK	986.58	Joback Method
cpg	1051.27	J/molxK	1022.22	Joback Method
cpg	1063.24	J/molxK	1057.86	Joback Method
dvisc	0.0011636	Paxs	470.00	Joback Method

dvisc	0.0005606	Paxs	532.34	Joback Method
dvisc	0.0003148	Paxs	594.67	Joback Method
dvisc	0.0001972	Paxs	657.00	Joback Method
dvisc	0.0001340	Paxs	719.34	Joback Method
dvisc	0.0000968	Paxs	781.67	Joback Method
dvisc	0.0000734	Paxs	844.01	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339740&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339740&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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