

# Butyric acid, 4-phenyl-, octadecyl ester

<b>Inchi:</b>	InChI=1S/C28H48O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-20-26-30-28(29)25-21-24
<b>InchiKey:</b>	VWJQEYHWHVGMNB-UHFFFAOYSA-N
<b>Formula:</b>	C28H48O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCOC(=O)CCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	416.68

## Physical Properties

Property code	Value	Unit	Source
gf	63.37	kJ/mol	Joback Method
hf	-629.52	kJ/mol	Joback Method
hfus	65.10	kJ/mol	Joback Method
hvap	89.35	kJ/mol	Joback Method
log10ws	-9.51		Crippen Method
logp	8.814		Crippen Method
mvol	389.060	ml/mol	McGowan Method
pc	802.97	kPa	Joback Method
rinpol	3129.00		NIST Webbook
rinpol	3129.00		NIST Webbook
tb	943.01	K	Joback Method
tc	1155.06	K	Joback Method
tf	503.90	K	Joback Method
vc	1.520	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1320.09	J/molxK	943.01	Joback Method
cpg	1340.86	J/molxK	978.35	Joback Method
cpg	1360.20	J/molxK	1013.69	Joback Method
cpg	1378.20	J/molxK	1049.03	Joback Method
cpg	1394.93	J/molxK	1084.38	Joback Method
cpg	1410.47	J/molxK	1119.72	Joback Method
cpg	1424.89	J/molxK	1155.06	Joback Method
dvisc	0.0005223	Paxs	503.90	Joback Method

dvisc	0.0002249	Paxs	577.08	Joback Method
dvisc	0.0001170	Paxs	650.27	Joback Method
dvisc	0.0000695	Paxs	723.45	Joback Method
dvisc	0.0000454	Paxs	796.64	Joback Method
dvisc	0.0000319	Paxs	869.82	Joback Method
dvisc	0.0000237	Paxs	943.01	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406188&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406188&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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