

# Fumaric acid, dec-4-enyl tetradecyl ester

<b>Inchi:</b>	InChI=1S/C28H50O4/c1-3-5-7-9-11-13-14-15-16-18-20-22-26-32-28(30)24-23-27(29)31-
<b>InchiKey:</b>	HRNSUFFPBMZNHF-BLXZHEFOSA-N
<b>Formula:</b>	C28H50O4
<b>SMILES:</b>	CCCCC=CCCCOC(=O)C=CC(=O)OCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	450.69

## Physical Properties

Property code	Value	Unit	Source
gf	-122.52	kJ/mol	Joback Method
hf	-876.41	kJ/mol	Joback Method
hfus	74.25	kJ/mol	Joback Method
hvap	96.15	kJ/mol	Joback Method
log10ws	-8.98		Crippen Method
logp	8.247		Crippen Method
mvol	411.660	ml/mol	McGowan Method
pc	726.53	kPa	Joback Method
rinpol	3171.00		NIST Webbook
rinpol	3171.00		NIST Webbook
tb	1000.94	K	Joback Method
tc	1236.34	K	Joback Method
tf	539.48	K	Joback Method
vc	1.611	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1418.29	J/molxK	1000.94	Joback Method
cpg	1439.94	J/molxK	1040.17	Joback Method
cpg	1460.04	J/molxK	1079.41	Joback Method
cpg	1478.68	J/molxK	1118.64	Joback Method
cpg	1495.98	J/molxK	1157.87	Joback Method
cpg	1512.04	J/molxK	1197.11	Joback Method
cpg	1526.96	J/molxK	1236.34	Joback Method
dvisc	0.0002721	Paxs	539.48	Joback Method

dvisc	0.0001173	Paxs	616.39	Joback Method
dvisc	0.0000610	Paxs	693.30	Joback Method
dvisc	0.0000361	Paxs	770.21	Joback Method
dvisc	0.0000235	Paxs	847.12	Joback Method
dvisc	0.0000165	Paxs	924.03	Joback Method
dvisc	0.0000122	Paxs	1000.94	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348948&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348948&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/87-856-6/Fumaric-acid-dec-4-enyl-tetradecyl-ester.pdf>

Generated by Cheméo on 2024-04-26 09:55:01.094309124 +0000 UTC m=+16414550.014886446.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.