

# 1,2-Cyclohexanedicarboxylic acid, 3,5-dimethylcyclohexyl dodecyl ester

Inchi:	InChI=1S/C28H50O4/c1-4-5-6-7-8-9-10-11-12-15-18-31-27(29)25-16-13-14-17-26(25)28
InchiKey:	FZIPNDRDSBTSIJ-UHFFFAOYSA-N
Formula:	C28H50O4
SMILES:	CCCCCCCCCCCCOC(=O)C1CCCCC1C(=O)OC1CC(C)CC(C)C1
Mol. weight [g/mol]:	450.69

## Physical Properties

Property code	Value	Unit	Source
gf	-257.19	kJ/mol	Joback Method
hf	-1063.23	kJ/mol	Joback Method
hfus	60.73	kJ/mol	Joback Method
hvap	96.17	kJ/mol	Joback Method
log10ws	-8.20		Crippen Method
logp	7.625		Crippen Method
mvol	398.540	ml/mol	McGowan Method
pc	814.93	kPa	Joback Method
rinpol	3078.00		NIST Webbook
rinpol	3078.00		NIST Webbook
tb	1017.71	K	Joback Method
tc	1246.07	K	Joback Method
tf	551.68	K	Joback Method
vc	1.514	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1478.72	J/molxK	1017.71	Joback Method
cpg	1497.73	J/molxK	1055.77	Joback Method
cpg	1514.34	J/molxK	1093.83	Joback Method
cpg	1528.60	J/molxK	1131.89	Joback Method
cpg	1540.55	J/molxK	1169.95	Joback Method
cpg	1550.22	J/molxK	1208.01	Joback Method
cpg	1557.68	J/molxK	1246.07	Joback Method
dvisc	0.0005758	Paxs	551.68	Joback Method

dvisc	0.0002821	Paxs	629.35	Joback Method
dvisc	0.0001617	Paxs	707.02	Joback Method
dvisc	0.0001035	Paxs	784.69	Joback Method
dvisc	0.0000718	Paxs	862.37	Joback Method
dvisc	0.0000529	Paxs	940.04	Joback Method
dvisc	0.0000408	Paxs	1017.71	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339854&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339854&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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