

# 1,2-Cyclohexanedicarboxylic acid, pentadecyl pentyl ester

Inchi:	InChI=1S/C28H52O4/c1-3-5-7-8-9-10-11-12-13-14-15-16-20-24-32-28(30)26-22-18-17-2
InchiKey:	UQUOSFZQLOYIML-UHFFFAOYSA-N
Formula:	C28H52O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C1CCCCC1C(=O)OCCCCC
Mol. weight [g/mol]:	452.71

## Physical Properties

Property code	Value	Unit	Source
gf	-266.22	kJ/mol	Joback Method
hf	-1076.87	kJ/mol	Joback Method
hfus	66.76	kJ/mol	Joback Method
hvap	96.35	kJ/mol	Joback Method
log10ws	-8.68		Crippen Method
logp	8.161		Crippen Method
mvol	409.400	ml/mol	McGowan Method
pc	753.91	kPa	Joback Method
rinpol	3180.00		NIST Webbook
rinpol	3180.00		NIST Webbook
tb	1007.50	K	Joback Method
tc	1238.90	K	Joback Method
tf	552.78	K	Joback Method
vc	1.583	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1474.90	J/molxK	1007.50	Joback Method
cpg	1495.49	J/molxK	1046.07	Joback Method
cpg	1513.96	J/molxK	1084.63	Joback Method
cpg	1530.36	J/molxK	1123.20	Joback Method
cpg	1544.76	J/molxK	1161.77	Joback Method
cpg	1557.22	J/molxK	1200.34	Joback Method
cpg	1567.79	J/molxK	1238.90	Joback Method
dvisc	0.0003982	Paxs	552.78	Joback Method

dvisc	0.0001832	Paxs	628.57	Joback Method
dvisc	0.0000996	Paxs	704.35	Joback Method
dvisc	0.0000610	Paxs	780.14	Joback Method
dvisc	0.0000407	Paxs	855.93	Joback Method
dvisc	0.0000290	Paxs	931.71	Joback Method
dvisc	0.0000218	Paxs	1007.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339477&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339477&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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