

# Fumaric acid, 2-methylpent-3-yl octadecyl ester

Inchi:	InChI=1S/C28H52O4/c1-5-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-24-31-27(29)22-
InchiKey:	NPLHVCCTMWOIOI-GHVJWSGMSA-N
Formula:	C28H52O4
SMILES:	CCCCCCCCCCCCCCCCCOC(=O)C=CC(=O)OC(CC)C(C)C
Mol. weight [g/mol]:	452.71

## Physical Properties

Property code	Value	Unit	Source
gf	-207.62	kJ/mol	Joback Method
hf	-1004.19	kJ/mol	Joback Method
hfus	67.01	kJ/mol	Joback Method
hvap	95.42	kJ/mol	Joback Method
log10ws	-8.99		Crippen Method
logp	8.325		Crippen Method
mvol	415.960	ml/mol	McGowan Method
pc	712.63	kPa	Joback Method
rinpol	3073.00		NIST Webbook
rinpol	3073.00		NIST Webbook
tb	995.90	K	Joback Method
tc	1229.46	K	Joback Method
tf	514.56	K	Joback Method
vc	1.619	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1447.04	J/molxK	995.90	Joback Method
cpg	1468.71	J/molxK	1034.83	Joback Method
cpg	1488.61	J/molxK	1073.75	Joback Method
cpg	1506.83	J/molxK	1112.68	Joback Method
cpg	1523.45	J/molxK	1151.60	Joback Method
cpg	1538.56	J/molxK	1190.53	Joback Method
cpg	1552.25	J/molxK	1229.46	Joback Method
dvisc	0.0003830	Paxs	514.56	Joback Method

dvisc	0.0001439	Paxs	594.78	Joback Method
dvisc	0.0000682	Paxs	675.01	Joback Method
dvisc	0.0000379	Paxs	755.23	Joback Method
dvisc	0.0000236	Paxs	835.45	Joback Method
dvisc	0.0000159	Paxs	915.68	Joback Method
dvisc	0.0000115	Paxs	995.90	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348774&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348774&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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