

# Citronellyl oleate

<b>Inchi:</b>	InChI=1S/C28H52O2/c1-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-23-28(29)30-25-24-2
<b>InchiKey:</b>	ZSJIOYGPFQLUAF-SEYXRHQNSA-N
<b>Formula:</b>	C28H52O2
<b>SMILES:</b>	CCCCCCCCC=CCCCCCCCC(=O)OCCC(C)CCC=C(C)C
<b>Mol. weight [g/mol]:</b>	420.71
<b>CAS:</b>	119742-35-7

## Physical Properties

Property code	Value	Unit	Source
gf	100.41	kJ/mol	Joback Method
hf	-646.68	kJ/mol	Joback Method
hfus	66.63	kJ/mol	Joback Method
hvap	86.69	kJ/mol	Joback Method
log10ws	-9.87		Crippen Method
logp	9.340		Crippen Method
mcvol	404.220	ml/mol	McGowan Method
pc	713.77	kPa	Joback Method
rinpol	2893.20		NIST Webbook
rinpol	2893.20		NIST Webbook
tb	924.09	K	Joback Method
tc	1133.11	K	Joback Method
tf	438.36	K	Joback Method
vc	1.583	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1362.36	J/mol×K	924.09	Joback Method
cpg	1385.20	J/mol×K	958.93	Joback Method
cpg	1406.77	J/mol×K	993.76	Joback Method
cpg	1427.15	J/mol×K	1028.60	Joback Method
cpg	1446.44	J/mol×K	1063.44	Joback Method
cpg	1464.74	J/mol×K	1098.28	Joback Method
cpg	1482.12	J/mol×K	1133.11	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C119742357&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C119742357&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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