

# Propane, 1,2,3-trichloro-2-(chloromethyl)-

<b>Other names:</b>	1,2,3-trichloro-2-(chloromethyl)propane
<b>Inchi:</b>	InChI=1S/C4H6Cl4/c5-1-4(8,2-6)3-7/h1-3H2
<b>InchiKey:</b>	CFLWPMRPCYBIIS-UHFFFAOYSA-N
<b>Formula:</b>	C4H6Cl4
<b>SMILES:</b>	CICC(Cl)(CCl)CCl
<b>Mol. weight [g/mol]:</b>	195.90
<b>CAS:</b>	18963-00-3

## Physical Properties

Property code	Value	Unit	Source
gf	-62.08	kJ/mol	Joback Method
hf	-197.60	kJ/mol	Joback Method
hfus	15.49	kJ/mol	Joback Method
hvap	40.74	kJ/mol	Joback Method
log10ws	-2.22		Crippen Method
logp	2.680		Crippen Method
mcvol	116.180	ml/mol	McGowan Method
pc	3337.38	kPa	Joback Method
rinpol	1105.00		NIST Webbook
rinpol	1105.00		NIST Webbook
tb	481.00 ± 6.00	K	NIST Webbook
tc	648.40	K	Joback Method
tf	256.94	K	Joback Method
vc	0.445	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	187.66	J/molxK	437.41	Joback Method
cpg	220.63	J/molxK	613.23	Joback Method
cpg	215.09	J/molxK	578.07	Joback Method
cpg	209.05	J/molxK	542.90	Joback Method
cpg	202.50	J/molxK	507.74	Joback Method
cpg	195.38	J/molxK	472.57	Joback Method

cpg	225.73	J/mol×K	648.40	Joback Method
dvisc	0.0004168	Paxs	437.41	Joback Method
dvisc	0.0005472	Paxs	407.33	Joback Method
dvisc	0.0007503	Paxs	377.25	Joback Method
dvisc	0.0010867	Paxs	347.18	Joback Method
dvisc	0.0016885	Paxs	317.10	Joback Method
dvisc	0.0028774	Paxs	287.02	Joback Method
dvisc	0.0055551	Paxs	256.94	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.48307e+01
Coeff. B	-4.13861e+03
Coeff. C	-7.57440e+01
Temperature range (K), min.	360.32
Temperature range (K), max.	514.72

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18963003&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18963003&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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