Propanoic acid, 2-oxo-, methyl ester

Other names: 2-oxopropanoic acid, methyl ester

Methylglyoxylic acid methyl ester

methyl 2-oxopropanoate

methyl pyruvate

pyruvic acid, methyl ester

Inchi: InChI=1S/C4H6O3/c1-3(5)4(6)7-2/h1-2H3
InchiKey: CWKLZLBVOJRSOM-UHFFFAOYSA-N

Formula: C4H6O3

SMILES: COC(=O)C(C)=O

Mol. weight [g/mol]: 102.09 CAS: 600-22-6

Physical Properties

Property code	Value	Unit	Source
gf	-380.04	kJ/mol	Joback Method
hf	-483.27	kJ/mol	Joback Method
hfus	10.50	kJ/mol	Joback Method
hvap	40.40	kJ/mol	Joback Method
ie	9.88	eV	NIST Webbook
log10ws	0.36		Crippen Method
logp	-0.252		Crippen Method
mcvol	76.230	ml/mol	McGowan Method
рс	4486.22	kPa	Joback Method
rinpol	701.00		NIST Webbook
rinpol	680.00		NIST Webbook
rinpol	710.00		NIST Webbook
rinpol	701.00		NIST Webbook
rinpol	701.00		NIST Webbook
rinpol	680.00		NIST Webbook
ripol	1217.00		NIST Webbook
ripol	1217.00		NIST Webbook
ripol	1180.00		NIST Webbook
ripol	1217.00		NIST Webbook
tb	408.70	K	NIST Webbook
tb	411.00 ± 6.00	K	NIST Webbook
tb	410.70 ± 0.60	K	NIST Webbook
tc	613.71	K	Joback Method

tf	256.93	K	Joback Method
VC	0.289	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source	
cpg	140.13	J/mol×K	421.08	Joback Method	
cpg	153.04	J/mol×K	485.29	Joback Method	
cpg	159.19	J/mol×K	517.39	Joback Method	
cpg	165.13	J/mol×K	549.50	Joback Method	
cpg	170.85	J/mol×K	581.60	Joback Method	
cpg	176.35	J/mol×K	613.71	Joback Method	
cpg	146.68	J/mol×K	453.18	Joback Method	
dvisc	0.0014709	Pa×s	284.29	Joback Method	
dvisc	0.0009841	Paxs	311.65	Joback Method	
dvisc	0.0007025	Pa×s	339.00	Joback Method	
dvisc	0.0005274	Paxs	366.36	Joback Method	
dvisc	0.0004120	Paxs	393.72	Joback Method	
dvisc	0.0023950	Paxs	256.93	Joback Method	
dvisc	0.0003324	Paxs	421.08	Joback Method	
pvap	1.23	kPa	307.80	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.92	kPa	302.90	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.71	kPa	298.90	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.53	kPa	294.90	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.43	kPa	291.00	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.36	kPa	289.20	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.30	kPa	287.00	Thermodynamic properties of pyruvic acid and its methyl ester	

pvap	0.29	kPa	286.20	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.30	kPa	286.20	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.25	kPa	283.80	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.23	kPa	282.40	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.19	kPa	280.90	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.17	kPa	279.40	Thermodynamic properties of pyruvic acid and its methyl ester	
pvap	0.18	kPa	279.40	Thermodynamic properties of pyruvic acid and its methyl ester	

Sources

NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C600226&Units=SI

Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Thermodynamic properties of pyruvic acid and its methyl ester: Joback Method:

https://www.doi.org/10.1016/j.tca.2018.05.009

https://en.wikipedia.org/wiki/Joback_method

McGowan Method: http://link.springer.com/article/10.1007/BF02311772

Legend

Ideal gas heat capacity cpg:

dvisc: Dynamic viscosity

Standard Gibbs free energy of formation gf: hf: Enthalpy of formation at standard conditions hfus: Enthalpy of fusion at standard conditions

Enthalpy of vaporization at standard conditions hvap:

ie: Ionization energy

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

pc: Critical Pressurepvap: Vapor pressure

rinpol: Non-polar retention indices

ripol: Polar retention indices

tb: Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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