

# 1,4-Dioxane-2,6-dione

<b>Other names:</b>	Diglycolic anhydride Acetic acid, oxydi-, cyclic anhydride p-Dioxane-2,6-dione Acetic acid, 2,2'-oxybis-, cyclic anhydride Diglycolic acid anhydride Oxydiacetic anhydride
<b>Inchi:</b>	InChI=1S/C4H4O4/c5-3-1-7-2-4(6)8-3/h1-2H2
<b>InchiKey:</b>	PIYNUZCGMLC XKJ-UHFFFAOYSA-N
<b>Formula:</b>	C4H4O4
<b>SMILES:</b>	O=C1COCC(=O)O1
<b>Mol. weight [g/mol]:</b>	116.07
<b>CAS:</b>	4480-83-5

## Physical Properties

Property code	Value	Unit	Source
gf	-402.46	kJ/mol	Joback Method
hf	-590.63	kJ/mol	Joback Method
hfus	11.86	kJ/mol	Joback Method
hvap	42.75	kJ/mol	Joback Method
log10ws	0.88		Crippen Method
logp	-0.914		Crippen Method
mcvol	71.240	ml/mol	McGowan Method
pc	5809.41	kPa	Joback Method
tb	513.70	K	NIST Webbook
tc	755.81	K	Joback Method
tf	336.04	K	Joback Method
vc	0.249	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	156.52	J/mol×K	504.68	Joback Method
cpg	166.76	J/mol×K	546.54	Joback Method
cpg	176.68	J/mol×K	588.39	Joback Method

cpg	186.20	J/mol×K	630.25	Joback Method
cpg	195.24	J/mol×K	672.10	Joback Method
cpg	203.75	J/mol×K	713.96	Joback Method
cpg	211.63	J/mol×K	755.81	Joback Method
hsubt	84.20 ± 1.10	kJ/mol	342.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4480835&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4480835&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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