

# 2-Propenamide, N,N-dimethyl-

<b>Other names:</b>	Acrylamide, N,N-dimethyl- N,N-Dimethylacrylamide Acylamide, N,N-dimethyl Dimethylamid kyseliny akrylove Propenamide, N,N-dimethyl-
<b>Inchi:</b>	InChI=1S/C5H9NO/c1-4-5(7)6(2)3/h4H,1H2,2-3H3
<b>InchiKey:</b>	YLGYACDQVQQZSW-UHFFFAOYSA-N
<b>Formula:</b>	C5H9NO
<b>SMILES:</b>	C=CC(=O)N(C)C
<b>Mol. weight [g/mol]:</b>	99.13
<b>CAS:</b>	2680-03-7

## Physical Properties

Property code	Value	Unit	Source
affp	904.30	kJ/mol	NIST Webbook
basg	873.40	kJ/mol	NIST Webbook
gf	60.92	kJ/mol	Joback Method
hf	-66.15	kJ/mol	Joback Method
hfus	12.05	kJ/mol	Joback Method
hvap	34.84	kJ/mol	Joback Method
log10ws	-0.11		Crippen Method
logp	0.261		Crippen Method
mvol	88.560	ml/mol	McGowan Method
pc	3930.78	kPa	Joback Method
tb	376.79	K	Joback Method
tc	557.36	K	Joback Method
tf	226.75	K	Joback Method
vc	0.321	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	154.25	J/mol×K	376.79	Joback Method
cpg	163.64	J/mol×K	406.88	Joback Method

cpg	172.58	J/mol×K	436.98	Joback Method
cpg	181.08	J/mol×K	467.07	Joback Method
cpg	189.16	J/mol×K	497.17	Joback Method
cpg	196.83	J/mol×K	527.26	Joback Method
cpg	204.11	J/mol×K	557.36	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	353.70	K	2.70	NIST Webbook
tbrp	355.00 ± 1.00	K	2.80	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2680037&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2680037&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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