

# 2-Methyl-4,5-dihydro-3-furanthiol

<b>Other names:</b>	Furan-3-thiol, 4,5-dihydro-2-methyl 2-Methyl-4,5(2H)-Furanthiol-3
<b>Inchi:</b>	InChI=1S/C5H8OS/c1-4-5(7)2-3-6-4/h7H,2-3H2,1H3
<b>InchiKey:</b>	IHRSRFITLMUQC-UHFFFAOYSA-N
<b>Formula:</b>	C5H8OS
<b>SMILES:</b>	CC1=C(S)CCO1
<b>Mol. weight [g/mol]:</b>	116.18

## Physical Properties

Property code	Value	Unit	Source
gf	-10.55	kJ/mol	Joback Method
hf	-124.39	kJ/mol	Joback Method
hfus	14.03	kJ/mol	Joback Method
hvap	40.15	kJ/mol	Joback Method
log10ws	-1.81		Crippen Method
logp	1.568		Crippen Method
mcvol	88.370	ml/mol	McGowan Method
pc	4931.51	kPa	Joback Method
rinpol	927.00		NIST Webbook
rinpol	966.00		NIST Webbook
rinpol	966.00		NIST Webbook
rinpol	919.00		NIST Webbook
rinpol	919.00		NIST Webbook
tb	432.68	K	Joback Method
tc	664.40	K	Joback Method
tf	250.08	K	Joback Method
vc	0.319	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	161.59	J/molxK	432.68	Joback Method
cpg	171.74	J/molxK	471.30	Joback Method
cpg	181.31	J/molxK	509.92	Joback Method

cpg	190.31	J/mol×K	548.54	Joback Method
cpg	198.77	J/mol×K	587.16	Joback Method
cpg	206.71	J/mol×K	625.78	Joback Method
cpg	214.15	J/mol×K	664.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R169096&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R169096&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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