

# Acetaldehyde, (1-methylethyl)hydrazone

<b>Other names:</b>	Acetaldehyde isopropylhydrazone Ethanal, isopropylhydrazone, syn (#2)
<b>Inchi:</b>	InChI=1S/C5H12N2/c1-4-6-7-5(2)3/h4-5,7H,1-3H3
<b>InchiKey:</b>	CHDZJJJPEGMSAM-UHFFFAOYSA-N
<b>Formula:</b>	C5H12N2
<b>SMILES:</b>	CC=NNC(C)C
<b>Mol. weight [g/mol]:</b>	100.16
<b>CAS:</b>	7422-89-1

## Physical Properties

Property code	Value	Unit	Source
hf	-16.12	kJ/mol	Joback Method
hvap	36.09	kJ/mol	Joback Method
log10ws	-1.37		Crippen Method
logp	0.990		Crippen Method
mcvol	96.970	ml/mol	McGowan Method
pc	3149.09	kPa	Joback Method
rinpol	770.00		NIST Webbook
rinpol	770.00		NIST Webbook
tb	440.21	K	Joback Method
tc	637.85	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7422891&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7422891&amp;Units=SI</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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