

# Acetonitrile, bromo-

<b>Other names:</b>	Bromoacetonitrile Bromomethyl cyanide Cyanomethyl bromide
<b>Inchi:</b>	InChI=1S/C2H2BrN/c3-1-2-4/h1H2
<b>InchiKey:</b>	REXUYBKPWIPONM-UHFFFAOYSA-N
<b>Formula:</b>	C2H2BrN
<b>SMILES:</b>	N#CCBr
<b>Mol. weight [g/mol]:</b>	119.95
<b>CAS:</b>	590-17-0

## Physical Properties

Property code	Value	Unit	Source
gf	113.46	kJ/mol	Joback Method
hf	106.60	kJ/mol	Joback Method
hfus	7.73	kJ/mol	Joback Method
hvap	36.96	kJ/mol	Joback Method
ie	11.28	eV	NIST Webbook
log10ws	-0.96		Crippen Method
logp	0.905		Crippen Method
mcvol	57.920	ml/mol	McGowan Method
pc	5359.18	kPa	Joback Method
rinpol	725.00		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	725.00		NIST Webbook
tb	413.40	K	Joback Method
tc	629.95	K	Joback Method
tf	237.09	K	Joback Method
vc	0.235	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	76.64	J/molxK	413.40	Joback Method
cpg	79.57	J/molxK	449.49	Joback Method

cpg	82.31	J/mol×K	485.58	Joback Method
cpg	84.88	J/mol×K	521.67	Joback Method
cpg	87.29	J/mol×K	557.76	Joback Method
cpg	89.55	J/mol×K	593.86	Joback Method
cpg	91.67	J/mol×K	629.95	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	334.20	K	3.20	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C590170&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C590170&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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