# Methanamine, N,N-dimethyl-, N-oxide

Other names: N,N-dimethylmethanamine N-oxide

> **TMAO** Triox

trimethylamine oxide trimethylamine, N-oxide

Inchi: InChI=1S/C3H9NO/c1-4(2,3)5/h1-3H3 UYPYRKYUKCHHIB-UHFFFAOYSA-N InchiKey:

Formula: C3H9NO

SMILES: C[N+](C)(C)[O-]

Mol. weight [g/mol]: 75.11

CAS: 1184-78-7

## **Physical Properties**

Property code	Value	Unit	Source
affp	983.20	kJ/mol	NIST Webbook
basg	953.50	kJ/mol	NIST Webbook
ie	$8.38 \pm 0.04$	eV	NIST Webbook
ie	8.27	eV	NIST Webbook
log10ws	0.34		Crippen Method
logp	0.190		Crippen Method
mcvol	68.980	ml/mol	McGowan Method

### Sources

**Crippen Method:** https://www.chemeo.com/doc/models/crippen\_log10ws

Compatible solutes: Thermodynamic https://www.doi.org/10.1016/j.fluid.2015.07.004 properties relevant for effective McGovian Methods: osmotic stress: http://link.springer.com/article/10.1007/BF02311772

**Crippen Method:** http://pubs.acs.org/doi/abs/10.1021/ci990307l Thermodynamics of the interactions of https://www.doi.org/10.1016/j.jct.2011.05.012 The modify names of the interactions of the interaction of t

http://webbook.nist.gov/cgi/cbook.cgi?ID=C1184787&Units=SI

https://www.doi.org/10.1016/j.jct.2011.12.029 https://www.doi.org/10.1021/je500977g

glycine peptides: interactions of some short peptides with the osmolyte trimethylamine Newsity and deluges in Properties of Anysous Relation metaling ethylamine N-Oxide in the Temperature Range from (278.15 to 323.15) K and at Pressures up to 100 MPa:

The hydration of the protein stabilizing  $\,$  https://www.doi.org/10.1016/j.jct.2013.01.023 agents: Trimethylamine-N-oxide,

agents: Trimethylamine-N-oxide, glycine and its N-methylderivatives The volumetric and compressibility

studies:

### Legend

**affp:** Proton affinity **basg:** Gas basicity

ie: Ionization energy

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

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