

# Rhodanine

<b>Other names:</b>	2,4-Thiazolidinedione, 2-thio- 2-Thio-4-ketothiazolidine 2-Thioxo-4-thiazolidinone 4-Oxo-2-thionothiazolidine 4-Oxo-2-thiothiazolidin 4-Oxo-2-thioxothiazoline 4-Thiazolidinone, 2-thioxo- 4-Thiazolidinone-2-thione NSC 1899 Rhodanic acid Rhodanin Rhodanic acid Rodanin USAF HA-2
<b>Inchi:</b>	InChI=1S/C3H3NOS2/c5-2-1-7-3(6)4-2/h1H2,(H,4,5,6)
<b>InchiKey:</b>	KIWUVOGUEXMXSV-UHFFFAOYSA-N
<b>Formula:</b>	C3H3NOS2
<b>SMILES:</b>	OC1=NC(=S)SC1
<b>Mol. weight [g/mol]:</b>	133.19
<b>CAS:</b>	141-84-4

## Physical Properties

Property code	Value	Unit	Source
gf	149.64	kJ/mol	Joback Method
hf	100.98	kJ/mol	Joback Method
hfus	16.14	kJ/mol	Joback Method
hvap	59.97	kJ/mol	Joback Method
log10ws	-1.77		Aqueous Solubility Prediction Method
log10ws	-1.77		Estimated Solubility Method
logp	0.975		Crippen Method
mcvol	82.220	ml/mol	McGowan Method
pc	7957.72	kPa	Joback Method
tb	558.48	K	Joback Method
tc	802.36	K	Joback Method
tf	442.65	K	Aqueous Solubility Prediction Method

vc

0.283

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	162.17	J/mol×K	558.48	Joback Method
cpg	168.35	J/mol×K	599.13	Joback Method
cpg	174.03	J/mol×K	639.77	Joback Method
cpg	179.24	J/mol×K	680.42	Joback Method
cpg	184.01	J/mol×K	721.06	Joback Method
cpg	188.38	J/mol×K	761.71	Joback Method
cpg	192.39	J/mol×K	802.36	Joback Method

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C141844&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**Estimated Solubility Method:**

[http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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