

# Glycidaldehyde

<b>Other names:</b>	2,3-epoxypropionaldehyde
<b>Inchi:</b>	InChI=1S/C3H4O2/c4-1-3-2-5-3/h1,3H,2H2
<b>InchiKey:</b>	IWYRWIUNAVNFPE-UHFFFAOYSA-N
<b>Formula:</b>	C3H4O2
<b>SMILES:</b>	O=CC1CO1
<b>Mol. weight [g/mol]:</b>	72.06
<b>CAS:</b>	765-34-4

## Physical Properties

Property code	Value	Unit	Source
gf	-150.51	kJ/mol	Joback Method
hf	-250.03	kJ/mol	Joback Method
hfus	11.93	kJ/mol	Joback Method
hvap	33.42	kJ/mol	Joback Method
log10ws	0.55		Crippen Method
logp	-0.416		Crippen Method
mcvol	49.710	ml/mol	McGowan Method
pc	5818.28	kPa	Joback Method
tb	350.39	K	Joback Method
tc	541.92	K	Joback Method
tf	210.08	K	Joback Method
vc	0.199	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	86.00	J/mol×K	350.39	Joback Method
cpg	92.66	J/mol×K	382.31	Joback Method
cpg	98.89	J/mol×K	414.23	Joback Method
cpg	104.71	J/mol×K	446.16	Joback Method
cpg	110.16	J/mol×K	478.08	Joback Method
cpg	115.25	J/mol×K	510.00	Joback Method
cpg	119.99	J/mol×K	541.92	Joback Method
dvisc	0.0011435	Paxs	210.08	Joback Method

dvisc	0.0009024	Paxs	233.46	Joback Method
dvisc	0.0007435	Paxs	256.85	Joback Method
dvisc	0.0006327	Paxs	280.24	Joback Method
dvisc	0.0005520	Paxs	303.62	Joback Method
dvisc	0.0004911	Paxs	327.00	Joback Method
dvisc	0.0004438	Paxs	350.39	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C765344&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C765344&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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