

# Octadeca-9,12,15-trienoic acid icosyl ester, Z,Z,Z

Inchi:	InChI=1S/C38H70O2/c1-3-5-7-9-11-13-15-17-19-20-21-23-25-27-29-31-33-35-37-40-38
InchiKey:	ZIFMUOBWYNUMBL-FLQWBAQUSA-N
Formula:	C38H70O2
SMILES:	CCC=CCC=CCC=CCCCCCCC(=O)OCCCCCCCCCCCCCCCCCCCC
Mol. weight [g/mol]:	558.96

## Physical Properties

Property code	Value	Unit	Source
gf	275.82	kJ/mol	Joback Method
hf	-720.79	kJ/mol	Joback Method
hfus	97.57	kJ/mol	Joback Method
hvap	109.21	kJ/mol	Joback Method
log10ws	-14.15		Crippen Method
logp	13.161		Crippen Method
mvol	540.820	ml/mol	McGowan Method
pc	459.71	kPa	Joback Method
rinpol	3926.60		NIST Webbook
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tb	1157.61	K	Joback Method
tc	1511.77	K	Joback Method
tf	574.94	K	Joback Method
vc	2.127	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2010.26	J/molxK	1157.61	Joback Method
cpg	2186.29	J/molxK	1452.75	Joback Method
cpg	2152.28	J/molxK	1393.72	Joback Method
cpg	2118.31	J/molxK	1334.69	Joback Method
cpg	2083.74	J/molxK	1275.66	Joback Method
cpg	2047.94	J/molxK	1216.64	Joback Method
cpg	2220.99	J/molxK	1511.77	Joback Method
dvisc	0.0000032	Paxs	1157.61	Joback Method

dvisc	0.0000044	Paxs	1060.50	Joback Method
dvisc	0.0000066	Paxs	963.39	Joback Method
dvisc	0.0000107	Paxs	866.28	Joback Method
dvisc	0.0000198	Paxs	769.16	Joback Method
dvisc	0.0000434	Paxs	672.05	Joback Method
dvisc	0.0001243	Paxs	574.94	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R437319&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R437319&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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