

# Hexadecanoic acid, 1,3-propanediyl ester

<b>Other names:</b>	Palmitic acid, trimethylene ester 1,3-Propanediol dipalmitate Hexadecanoic acid, 1,1'-(1,3-propanediyl) ester
<b>Inchi:</b>	InChI=1S/C35H68O4/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-30-34(36)38-32-29-33-39-
<b>InchiKey:</b>	FBFGIPIHQQUQF-UHFFFAOYSA-N
<b>Formula:</b>	C35H68O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCC(=O)OCCOC(=O)CCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	552.91
<b>CAS:</b>	818-21-3

## Physical Properties

Property code	Value	Unit	Source
gf	-224.02	kJ/mol	Joback Method
hf	-1255.33	kJ/mol	Joback Method
hfus	91.98	kJ/mol	Joback Method
hvap	111.82	kJ/mol	Joback Method
log10ws	-12.20		Crippen Method
logp	11.426		Crippen Method
mcvol	518.890	ml/mol	McGowan Method
pc	496.91	kPa	Joback Method
tb	1152.78	K	Joback Method
tc	1519.56	K	Joback Method
tf	628.53	K	Joback Method
vc	2.043	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2049.57	J/molxK	1519.56	Joback Method
cpg	1935.35	J/molxK	1152.78	Joback Method
cpg	1964.52	J/molxK	1213.91	Joback Method
cpg	1989.12	J/molxK	1275.04	Joback Method
cpg	2009.54	J/molxK	1336.17	Joback Method
cpg	2026.17	J/molxK	1397.30	Joback Method

cpg	2039.39	J/mol×K	1458.43	Joback Method
dvisc	0.0000050	Paxs	1152.78	Joback Method
dvisc	0.0001166	Paxs	628.53	Joback Method
dvisc	0.0000502	Paxs	715.90	Joback Method
dvisc	0.0000259	Paxs	803.28	Joback Method
dvisc	0.0000153	Paxs	890.65	Joback Method
dvisc	0.0000099	Paxs	978.03	Joback Method
dvisc	0.0000069	Paxs	1065.40	Joback Method
hfust	133.00	kJ/mol	329.80	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C818213&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C818213&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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