

# Imidosulfurous difluoride, (fluorocarbonyl)-

<b>Other names:</b>	Imidosulfurous difluoride, (fluoroformyl)- N-Fluoroformyl iminosulfur difluoride S,S-difluoro-N-(fluoroformyl)-sulfimine
<b>Inchi:</b>	InChI=1S/CF3NOS/c2-1(6)5-7(3)4
<b>InchiKey:</b>	KVNVOUIZJYPKAF-UHFFFAOYSA-N
<b>Formula:</b>	CF3NOS
<b>SMILES:</b>	O=C(F)N=S(F)F
<b>Mol. weight [g/mol]:</b>	131.08
<b>CAS:</b>	3855-41-2

## Physical Properties

Property code	Value	Unit	Source
hf	-755.21	kJ/mol	Joback Method
hvap	32.41	kJ/mol	Joback Method
log10ws	-0.65		Crippen Method
logp	1.647		Crippen Method
mcvol	58.160	ml/mol	McGowan Method
pc	4194.74	kPa	Joback Method
tb	418.04	K	Joback Method
tc	617.42	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	37.30	kJ/mol	271.50	NIST Webbook

## Sources

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C3855412&Units=SI>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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