

# 1,2-Cyclohexanedicarboxylic acid, 2-cyclohexylethyl tetradecyl ester

**Inchi:** InChI=1S/C30H54O4/c1-2-3-4-5-6-7-8-9-10-11-12-18-24-33-29(31)27-21-16-17-22-28(29)  
**InchiKey:** JOKNBIUQXMHWFA-UHFFFAOYSA-N  
**Formula:** C30H54O4  
**SMILES:** CCCCCCCCCCCCCOC(=O)C1CCCCC1C(=O)OCCC1CCCCC1  
**Mol. weight [g/mol]:** 478.75

## Physical Properties

Property code	Value	Unit	Source
gf	-224.93	kJ/mol	Joback Method
hf	-1063.83	kJ/mol	Joback Method
hfus	63.77	kJ/mol	Joback Method
hvap	101.23	kJ/mol	Joback Method
log10ws	-9.17		Crippen Method
logp	8.551		Crippen Method
mvol	426.720	ml/mol	McGowan Method
pc	759.32	kPa	Joback Method
rinpol	3431.00		NIST Webbook
rinpol	3431.00		NIST Webbook
tb	1072.81	K	Joback Method
tc	1317.67	K	Joback Method
tf	582.70	K	Joback Method
vc	1.629	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1598.70	J/molxK	1072.81	Joback Method
cpg	1667.56	J/molxK	1276.86	Joback Method
cpg	1658.49	J/molxK	1236.05	Joback Method
cpg	1647.15	J/molxK	1195.24	Joback Method
cpg	1633.46	J/molxK	1154.43	Joback Method
cpg	1617.34	J/molxK	1113.62	Joback Method
cpg	1674.45	J/molxK	1317.67	Joback Method
dvisc	0.0000163	Paxs	1072.81	Joback Method

dvisc	0.0000219	Paxs	991.12	Joback Method
dvisc	0.0000311	Paxs	909.44	Joback Method
dvisc	0.0000474	Paxs	827.75	Joback Method
dvisc	0.0000791	Paxs	746.07	Joback Method
dvisc	0.0001497	Paxs	664.38	Joback Method
dvisc	0.0003389	Paxs	582.70	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339733&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339733&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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