

# 1,2-Cyclohexanedicarboxylic acid, heptyl octyl ester

Inchi:	InChI=1S/C23H42O4/c1-3-5-7-9-11-15-19-27-23(25)21-17-13-12-16-20(21)22(24)26-18-
InchiKey:	STURPVSRWWYUMZ-UHFFFAOYSA-N
Formula:	C23H42O4
SMILES:	CCCCCCCCOC(=O)C1CCCCC1C(=O)OCCCCCCC
Mol. weight [g/mol]:	382.58

## Physical Properties

Property code	Value	Unit	Source
gf	-308.32	kJ/mol	Joback Method
hf	-973.67	kJ/mol	Joback Method
hfus	53.81	kJ/mol	Joback Method
hvap	85.22	kJ/mol	Joback Method
log10ws	-6.59		Crippen Method
logp	6.210		Crippen Method
mvol	338.950	ml/mol	McGowan Method
pc	1000.18	kPa	Joback Method
rinpol	2623.00		NIST Webbook
rinpol	2623.00		NIST Webbook
tb	893.10	K	Joback Method
tc	1095.01	K	Joback Method
tf	496.43	K	Joback Method
vc	1.304	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1155.63	J/molxK	893.10	Joback Method
cpg	1175.24	J/molxK	926.75	Joback Method
cpg	1193.32	J/molxK	960.40	Joback Method
cpg	1209.91	J/molxK	994.05	Joback Method
cpg	1225.04	J/molxK	1027.71	Joback Method
cpg	1238.73	J/molxK	1061.36	Joback Method
cpg	1251.01	J/molxK	1095.01	Joback Method
dvisc	0.0007325	Paxs	496.43	Joback Method

dvisc	0.0003525	Paxs	562.54	Joback Method
dvisc	0.0001979	Paxs	628.65	Joback Method
dvisc	0.0001240	Paxs	694.76	Joback Method
dvisc	0.0000842	Paxs	760.88	Joback Method
dvisc	0.0000609	Paxs	826.99	Joback Method
dvisc	0.0000462	Paxs	893.10	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339539&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339539&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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